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CHEMISTRY DIVISION
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Report on

AN EMULSTION PASTE SYSTEM FOR DECONTAMINATION OF VESICANTS

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ABSTRACT

A new decontamination system for vesicants has been developed, which consists of a chloroamide dispersed in an emulsion of water in Perclene (tetrachloroethylene). This paste system was investigated in detail to determine the effect of varying its composition and the conditions of use on its effectiveness as a decontaminant. The improved NDRC potassium oleate paste was examined and both paste systems compared to TCE/RH-195 solution. TCE/RH-195 was found to be the most efficient and the easiest to use of the three systems for the decontamination of H in deck paint.

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INTRODUCTION

Authorization

1. This work was authorized under Bureau of Ships Project No. 397/44. Problems proposed for study were given in Bureau of Ships letter S-S77-2(Dz) Serial 811 dated 17 December 1940.

Statement of the Problem

2. This is a continuation of the study of the decontamination of C. W. Agents, with special emphasis on the development of a paste system containing a large proportion of water.

Known Facts Bearing on the Problem

3. The present Navy standard decontamination system for shipboard use is a 10% solution of RH-195 in tetrachloroethane (TCE/RH-195). The objectionable characteristics of this solution have been discussed in previous reports from this Laboratory. They include its toxicity, corrosiveness and injurious effect on various materials such as paint, clear plastics, rubber and cellulose acetate butyrate doped fabrics.

4. A paste system developed by the NDRC, which consists essentially of Perclene (tetrachloroethylene) potassium oleate and S-461 or S-210, has some advantages over the TCE/RH-195 system. This Perclene/potassium oleate/chloroamide paste was discussed in detail in NRL Report No. P-2211. This report also mentioned the preliminary work at this Laboratory on a promising emulsion paste containing Perclene, an emulsifying agent, a chloroamide and a large proportion of water.

Theoretical Considerations

5. For shipboard use it would be desirable to have a decontamination system requiring the storage of as small an amount of materials as possible. One means of accomplishing this might be the substitution of water for all or part of the solvent, retaining a chloroamide as the active ingredient. Preliminary examination of water systems indicated that an emulsion applied as a paste (too fluid an emulsion would run off vertical surfaces quickly) offered promise. The emulsion should contain a paint penetrant, preferably one which will dissolve certain chloroamides to some extent.

6. To decontaminate most efficiently, it was felt that the emulsion should be of the water-in-oil type so that the chloroamide-bearing paint penetrant could come in contact with the contaminated surfaces. Furthermore, previous work indicated that some water-in-oil emulsions would stick to vertical surfaces, whereas oil-in-water emulsions tended to run off quickly. The emulsion should be fairly stable so that it will not break readily when applied to a vertical surface and allow the paste to run off.

7. In the preliminary work toward development of an emulsion paste a suitable water-insoluble solvent was needed. Perclene (tetrachloroethylene) was selected as the most likely solvent to use on the basis of certain desirable characteristics found in earlier work. One disadvantage of Perclene as compared to TCE is that Perclene is much more volatile than TCE at ordinary temperatures. It was felt that incorporation in an emulsion would increase the contact time of the Perclene with the painted surfaces, thereby increasing the efficiency of the solvent.

Previous Work Done at this Laboratory

8. Previous work done at this Laboratory related to this problem has been presented in NRL Report No. P-1944, "The Use of RH-195 for the Decontamination of HS and M-1", dated 8 October 1942; NRL Report No. P-2125, "A Study of Perclene (Tetrachloroethylene) as a Solvent for use in the Decontamination of Airplanes", dated 29 July 1943; and NRL Report No. P-2211, "Chloroamide Paste Systems for Decontamination of Vesicants" dated 31 December 1943.

EXPERIMENTAL

PART I. DEVELOPMENT OF AN EMULSION PASTE FOR DECONTAMINATION

A. Screening Tests of Emulsifying Agents

9. It was desired to make an emulsion of water-in-Perclene into which a chloroamide could be incorporated. Therefore, an arbitrary series of requirements for candidate emulsifying agents was set up as follows:

(1) The emulsifying agent must make a water-in-oil type of emulsion in which Perclene is the continuous phase.

(2) The emulsifying agent must make an emulsion which will adhere well to vertical surfaces.

(3) The emulsion must be stable when chloroamides are dispersed in it.

(4) The chloroamide-containing emulsion must be prepared easily by stirring the ingredients by hand with a paddle.

(5) The emulsifying agent should preferably be soluble in Perclene, and this solution should be stable under various conditions of storage.

(6) A somewhat viscous emulsion would be desirable because it would require less chloroamide powder to make a suitably thick paste.

Much of the preliminary work consisted of screening the emulsifying agents on the basis of these requirements. Additional factors were expected to influence the selection of the best emulsifying agent from those which satisfied these six requirements.

(1) Preliminary Tests

10. Over 100 surface-active agents were tested as follows: One gram of the surface-active agent was mixed with 20 grams of Perclene and the mixture examined for solubility of the agent in Perclene. Water was then added in 10 ml. increments, usually up to a total of 50 ml., and the mixture stirred after each addition with a Hamilton Beach high-speed stirrer. The mixture was then examined for (a) emulsion formation, (b) type of emulsion, (c) stability of the emulsion, and (d) thickening action. The results are tabulated in Appendix A, Table I.

11. Those agents which gave no emulsion were dropped, as were a number which gave O/W emulsions. None of the O/W type showed evidence of appreciably thickening the emulsions. Of those agents which gave water-in-oil emulsions, a number were eliminated because the emulsions broke quickly.

(2) Preparation of Hand-Stirred Emulsions Containing Chloroamides

12. The 25 agents which survived the screening tests were examined further by a procedure similar to that described above, except that hand stirring only was used. At the same time emulsions were prepared with the addition of S-461 powder. In no case did the addition of S-461 decrease the ease of emulsification or stability of the emulsion. In some cases the addition of S-461 improved the speed of emulsification. From observations based on these experiments, the emulsifying agents were divided arbitrarily into three classes: most promising, intermediate and least promising. This was the basis for the classification shown in Appendix A, Table II. In the most promising class were eight emulsifying agents, all of which were long chain fatty acid esters of sorbitan, mannitan or mannide, and all were products made by the Atlas Powder Co.

(3) Decontamination Tests on Deck Paint

13. Emulsion pastes were made up using the eight most promising emulsifying agents with S-461, RH-195 and S-210, according to the following formula:

20 g. Perclene
1 g. Emulsifying Agent
30 g. Tap water
5 g. S-461, RH-195 or S-210

14. Decontamination tests were made in the following manner: Two-inch square steel panels previously painted with zinc chromate primer and two coats of deck paint were contaminated with 2 drops (0.05 g.) of H spread on and allowed to stand for one hour. The decontamination paste was then liberally applied and left for one hour, then washed with water under the tap, the panels blotted and dried and tested at 35°C with Congo Red-S-328 test paper, in conjunction with sodium carbonate filter paper. The paper test times were recorded in minutes and all tests which were still negative after 45 minutes were recorded as negative.

15. The results obtained for the decontamination of deck paint with the emulsion pastes are given in Appendix A, Table III. Satisfactory decontamination was obtained with all pastes except S-461 and RH-195 with mannitan triricinoleate, RH-195 with mannitan diricinoleate and S-210 with Span 40 and Span 60.

In general, pastes made with S-461 were most viscous, less viscous with S-210, and the RH-195 pastes were thin. Mannitan diricinoleate and mannitan triricinoleate emulsions were the least satisfactory with all three chloroamides. They gave poor emulsions which broke and dried out on the panels, especially with RH-195. Arlacel A was somewhat better, but not as good as the remaining five emulsifying agents. Span 40, Span 60 and mannitan monolaurate made the thickest pastes when used in the above formulation. It was later learned that the manufacture of derivatives of mannitol had been suspended. The mannitan derivatives were in general no better than those of sorbitan. Therefore, only a few additional experiments were made with mannitan esters.

B. Factors Affecting the Efficiency of Emulsion Pastes for the Decontamination of H.

16. Because of the promise shown by the emulsion pastes in preliminary tests, a study was made of the effect of changing the proportions and ingredients. At the same time the decontamination efficiency for H under various conditions was studied.

(1) Effect of Varying the Perclene/Water Ratio

17. The first tests were made with a Perclene/water ratio of 20/30 by weight. In another experiment emulsion pastes of S-461 were prepared in which the amount of water was increased to give a ratio of 20/50. This data is included in Appendix A, Table III. These pastes decontaminated about as well as did similar 20/30 pastes. The increased water content made the pastes more viscous.

18. In another experiment four S-461 emulsion pastes were prepared using a 20/80 ratio of Perclene/water. The emulsifying agents used were Arlacel C, Span 80, Span 60 and Span 40. These pastes decontaminated H satisfactorily as shown in Appendix A, Table III. However, increasing the Perclene/water ratio to 20/80 made it more difficult to prepare the emulsions by hand. The emulsions were less stable and allowed water to separate.

19. The Perclene/water ratio was varied also in other experiments. It was found that a smaller proportion of water gave less viscous emulsions which tended to run off vertical surfaces. A larger proportion of water gave unstable emulsions which were more difficult to prepare by hand stirring. As a result of these experiments a 20/50 ratio of Perclene/water by weight was chosen as most likely to give good results.

(2) Effect of Varying the Amount of Emulsifying Agent

20. Pastes with a Perclene/water ratio of 20/80 were prepared using 5% S-461, RH-195 or S-210. The amount of emulsifying agent, Span 40 or Span 80, was varied from 1%, 5% to 10% of the weight of Perclene. These pastes were tested for decontamination of H in deck paint. All the S-461 and RH-195 pastes decontaminated H completely. The data for the S-210 pastes is given in Appendix A, Table IV.

21. There was no significant difference in decontamination efficiency due to variation in percentage of emulsifying agent. The emulsions made using 1% Span 40 or Span 80 were difficult to prepare and broke quickly. The 5% and 10% emulsions were satisfactory except for the RH-195 pastes which were less stable. The 10% emulsifying agent emulsions were the most viscous, the 5% somewhat less viscous and the 1% were quite thin. This effect was noted in other experiments in which the amount of emulsifying agent was varied from 5% to 10% and the Perclene/water ratio was 20/30, 20/50, 20/65 and 20/80. On the basis of these and other experiments, a 10% solution of emulsifying agent in Perclene was considered to be the most satisfactory.

(3) Choice of Chloroamide

22. As a result of a number of preliminary experiments some of which have already been described, the characteristics of RH-195, S-461 and S-210 in the emulsion pastes had been noted. The RH-195 and S-210 used had been micronized. The emulsions containing S-461 were stable and the decontamination efficiency was good. S-210 emulsion pastes were not as viscous as those made with S-461 and did not decontaminate quite as well, although they were stable. RH-195 emulsion pastes were thin, unstable with "leaking" of water, and were not suited for application to vertical surfaces. When applied to horizontal panels decontamination was usually obtained.

23. Of the chloroamides listed above, S-461 gave the best emulsion pastes However, S-461 is easily ignited and decomposes spontaneously after ignition. A number of other chloroamides and mixtures of S-461 with materials designed to inhibit the decomposition of S-461 on ignition were tested for use in emulsion pastes. Attempts were made to prepare 20/50 type emulsions containing 5% by weight of Span 20 (sorbitan monolaurate) or Span 80 (sorbitan monooleate) in Perclene and containing the chloroamides or mixtures listed below in the amount of 10% of the final paste:

(a) S-330 (micronized) made good stiff emulsions with both Spans, with no sign of breaking after 3 hours.

(b) S-222 made good, fairly stiff emulsions which were still quite good after 3 hours.

(c) S-300 emulsions made with Span 80 broke within a few minutes. They were readily re-emulsified, but were thin and broke quickly. S-300 emulsions made with Span 20 were much better than with Span 80, although they "leaked" some after an hour and were definitely thinner after three hours.

(d) S-145 emulsions began to break after about one hour. Span 20 was better than Span 80.

(e) RH-851 emulsions with Span 20 had broken slightly within an hour; with Span 80 they broke quickly and became quite thin.

(f) S-436 reacted vigorously with Span 80 with the evolution of heat. However, stable viscous emulsions containing S-436 were prepared using Span 20.

(g) S-461/Infusorial earth mixtures in Perclene prevented emulsification.

(h) Infusorial earth prevented emulsification.

(i) S-461/Bentonite mixtures prevented emulsification.

(j) Bentonite prevented emulsification.

24. Of those listed above, S-330, S-222 and S-436 made the best emulsions and were tested for the decontamination of H in deck paint. The results are given in Appendix A, Table V. Except for S-330 pastes made with Span 80, none of these pasts decontaminated satisfactorily. The pastes made with these chloroamides "skated" over the panels where liquid mustard was still standing and did not mix well with the H. This is in contrast to S-461, S-210 and RH-195 pastes, which seemed to absorb liquid H readily, adhering as well to the places where liquid H was still standing as where liquid H was not present.

(4) Substitution of Sea Water for Tap Water

25. The emulsion pastes studied thus far were made with tap water. Because it would be desirable to be able to make these pastes with sea water as well as fresh water, some emulsions were prepared using a "synthetic" sea water. Emulsion pastes prepared with sea water, Perclene, S-461 and mannitan monolaurate, Span 40, Span 60 or Span 80 were thin and broke quickly. They were unsatisfactory for application to vertical surfaces, but when applied to horizontal deck paint panels decontaminated H satisfactorily. Sea water emulsion pastes made with S-436 had fairly good physical properties, but decontaminated H poorly as did similar S-436 emulsions made with tap water.

26. It was found possible to make fairly satisfactory emulsion pastes with sea water by adding polybutene or other stabilizers in addition to the emulsifying agent. In the event that emulsion pastes should be considered practical for use, suitable modification could doubtless be made to allow the use of sea water in the formulations.

(5) Length of Time of Application of Emulsion Paste

27. All previous trials for decontamination of H in deck paint were made by leaving the paste on for one hour. S-210 pastes using both Span 20 and Span 80 were used to decontaminate H in deck paint, the paste being left on for 20, 40 or 60 minutes. The results of these tests are given in Appendix A, Table VI. It is apparent that 20 minutes is not sufficient time and 60 minutes gives somewhat better results than 40 minutes. On the basis of these results it was judged that a 60-minute application should be used.

(6) Effect of Temperature on Preparation and Decontamination

28. Pastes were made using Span 20 or Span 80, Perclene, S-461 and water as follows: (1) ice water, (2) tap water at room temperature, and (3) tap water at 70°C. Water at all three temperatures was easily emulsified. There was a slight difference in the viscosity of the pastes, the hot water pastes being less viscous than the cold. The emulsion pastes made with ice water were kept at 4°C, those made with room temperature water were kept at room temperature, and those made with 70°C water were kept at 45°C for one-half hour. There was no observable change in these pastes during that time.

29. The decontamination of H in deck paint at low temperatures was tested as follows: Deck paint panels were contaminated at room temperature and left for 45 minutes, then placed outdoors (11°C) for 15 minutes and finally kept at 3°C for 15 minutes before the paste was applied. The pastes were 20/50 type with S-210 and were prepared from (1) 5% Span 20 solution in Perclene kept at 3°C for 6 days, (2) 10% Span 20 solution in Perclene kept at 3°C for 10 days, and (3) 5% Span 80 solution in Perclene kept at 3°C for 18 days. The water used was at 3°C. There were no significant differences in the ease of preparation or stability of these pastes compared to pastes prepared from similar materials previously brought to room temperature. The pastes prepared at 3°C were applied to the H-contaminated panels and left at 3°C for one hour, then the pastes were removed and the panels tested at 35°C. All results showed complete decontamination under these conditions.

(7) Sequence of Addition of Ingredients

30. It was found that S-461 emulsion pastes could not be prepared by first mixing the S-461 with water, then adding the Perclene solution of the emulsifying agent. Emulsifying agents tried were Span 40, Span 60 and Span 80. Emulsion pastes containing S-210 and RH-195 could be prepared by mixing the chloroamide and water first. Emulsion pastes of all three chloroamides were successfully prepared by adding the ingredients in the following order: Perclene solution of Span, water, chloroamide; or Perclene solution of Span, chloroamide, water. The latter procedure was the easiest to use. It was concluded that the best method of preparing the emulsion pastes was to wet out the chloroamide in the Perclene solution of the emulsifying agent first, and then add the water with vigorous stirring.

(8) The Use of Other Solvents

31. Although Perclene was a desirable decontamination solvent in many ways, certain tests indicated that Perclene was more volatile than desired, even when used in the emulsion pastes. For this reason several other solvents were examined.

32. The characteristics of the solvents were as follows:

Hexachlorobutadiene (Hooker Electrochemical Company) is non-inflammable, b.p. 210-220°C., m.p. -19 to -22°C., and is not easily hydrolyzed by water or mild alkalies. Emulsion pastes made with this solvent were physically satisfactory and did not harm deck paint appreciably.

Hexachloropropene is non-inflammable and has a high boiling point and a low melting point. It likewise was readily emulsified and did not harm deck paint.

Chloropropane Liquid 170 (Hooker Electrochemical Company) is non-inflammable and has a boiling range of 160° to 260°C and does not freeze but becomes quite viscous below -50°C. Good emulsion pastes were made with this solvent which did not harm deck paint.

Butyl laurate is a relatively non-inflammable, high-boiling solvent with a melting point of less than -10°C. Emulsions made with this solvent were not very stable, although there was no indication of a reaction between chloroamide and solvent.

33. The results of decontamination tests with pastes made using these solvents are given in Appendix A, Table VII. None of the emulsion pastes containing these solvents decontaminated deck painted panels very well. Slurries of chloroamides mixed with these solvents did not decontaminate well either. Determinations of the solubility of chloroamides in hexachlorobutadiene and hexachloropropene (Table VIII) indicated that the failure to decontaminate was probably not due to lack of solubility of the chloroamides in these solvents.

34. Several other solvents were tried, but they did not perform well in the pastes. Tetrachloroethane made good emulsion pastes, which, however, completely stripped the paint from the test panels, so that no tests were made for completeness of decontamination. It was concluded that Perclene was the best solvent of those tried for use in these emulsion pastes in spite of its volatility.

(9) Conclusion

35. As a result of the studies outlined above, the following ingredients, proportions and procedure were selected for the preparation of the emulsion type decontamination pastes:

Ingredient	Parts by Weight
Perclene	18
Span 20 (sorbitan monolaurate) or Span 80 (sorbitan monooleate)	2
S-461 or S-210	7
Water	50

The paste is prepared by mixing the chloroamide with the Perclene solution of the emulsifying agent and then adding the water with stirring. This paste is then applied to the contaminated surface and left for one hour, after which it is flushed off with a stream of water, using brushes if necessary.

C. Decontamination of H in Various Materials

(1) Wet Deck Paint

36. In the tests of the emulsion pastes for the decontamination of H described thus far, steel panels painted with deck paint were used exclusively.

The panels used were dry. Further tests were made to determine whether the emulsion pastes would decontaminate deck paint wet with fresh water or sea water. After the H had been left on the panels for one hour, they were wet down with fresh water and sea water respectively, and then the emulsion paste was applied. The pastes adhered very well to the wet panels and did not harm the paint. The paper test times are given in Appendix A, Table IX. It is evident that decontamination of H in deck paint was successful under these conditions also. It may be noted here that the emulsion pastes did not have any harmful effect on deck paint, wet or dry.

(2) Airplane Paints

37. Decontamination tests were made for H on both M485 and DuPont 71 Line lacquers over zinc chromate primer on steel. These panels were contaminated with H and decontaminated with the pastes in the same manner as were the deck paint panels. The results of the tests on M485 lacquer are given in Appendix A, Table X. The M485 paint was considerably softened so that some of it was removed during the removal of the paste. The DuPont 71 Line panels were decontaminated completely by the emulsion paste treatment even after 20 minute application, and were apparently unharmed by the treatment.

(3) Doped Fabrics

38. S-461 and S-210 emulsion pastes using Span 20 or Span 80 were used to decontaminate H in doped fabrics, both cellulose nitrate and cellulose acetate butyrate types. The results are given in Appendix A, Table XI. The cellulose nitrate doped fabrics were decontaminated but the cellulose acetate butyrate fabrics were not. No harmful effect was noted on either type of coating. The experiment was repeated on cellulose acetate butyrate panels, except that the pastes were left on for two hours. Slightly better results were obtained. To determine the effect on the tensile strength of fabrics contaminated with H and decontaminated with the pastes, or fabrics treated with the pastes only, measurements were made on the series listed in Appendix A, Table XII. Each contamination and decontamination treatment was made three times at three day intervals. Where H was used, it was left at least one hour before applying the pastes. The tensile strength measurements (one-inch grab test) were made after two weeks outdoor exposure following the last treatment. Each figure given in the table is the average of four measurements. The data show that tensile strength of these fabrics was practically unaffected by the emulsion pastes applied with or without previous contamination with H.

(4) Clear Plastics

39. Decontamination of H in Lucite, Plexiglas and cellulose acetate sheet was attempted with pastes of the 20/50 Perclene/water type made with S-210 and S-461. Span 20 and Span 80 were the emulsifying agents. Two drops (0.05 g.) of H was left on 2" x 2" plastic panels for one hour and the paste applied and left for one hour. All the panels were completely decontaminated except one Plexiglas panel, which gave a negative test after 24 hours. No damage to the plastics was apparent except that caused by the exposure to liquid H.

D. Decontamination of Other C. W. Agents

(1) Decontamination of HV

40. Emulsion pastes of the 20/30 Perclene/water type made with S-461 using 5 different emulsifying agents were used to decontaminate HV (H containing 8% chlorinated rubber) on deck paint panels. The HV was left on for one hour before application of the paste. The results of the paper tests are given in Appendix A, Table XIII. The decontamination was good and the paint was not harmed.

(2) Decontamination of L

41. S-461 emulsion pastes of the 20/30 Perclene/water type, using 5 different emulsifying agents were used to decontaminate L on deck paint panels. The L was allowed to remain for one hour before decontamination was attempted. DT paper was used to detect any residual L after decontamination. If no positive test was obtained in 120 minutes at 35°C the test was recorded as negative. The results are given in Appendix A, Table XIII. In all cases decontamination was complete. The panels were bleached and pin-point blisters were numerous on each panel wherever L had been spread. This is the usual effect of L on deck paint.

(3) Decontamination of Nitrogen Mustards

(a) HN-3

42. Deck paint panels (2" x 2") were each contaminated with 0.05 g. of HN-3. After one hour, decontamination was attempted by several different methods. Tests for residual HN-3 were made with sodium carbonate papers spotted with DB-3 reagent used in conjunction with a filter of sodium carbonate paper. If no positive test was obtained within 30 minutes, the tests were reported as negative. The results are given in Appendix A, Table XIV. In addition to emulsion pastes of S-210 and S-461, decontamination was attempted by spraying the panels with a solution of RH-195 or S-436 in TCE. Comparison with the blank shows that all the methods used were helpful in the removal of HN-3 but that the spray of TCE/S-436 was outstanding in its beneficial effect.

(b) HN-1

43. Tests similar to those for HN-3 were made using HN-1 as the contaminant. The results are given in Appendix A, Table XV. The S-436 emulsion paste was the best decontaminating system and the spray of TCE/S-436 was good. None of the other methods were very effective, although the surface contamination was removed.

(4) Decontamination of Lachrymators

44. Some attempts were made to decontaminate CN and BBC in deck paint. The emulsion pastes were not very effective in the removal of either of these lachrymators. The Perclene/potassium oleate/S-461 paste discussed in a previous report was very effective in removing CN but did not decontaminate BBC completely.

E. Other Considerations Regarding Emulsion Pastes

45. The emulsion pastes were examined in regard to several other properties of importance for a decontaminating system.

(1) Storage of Perclene/Emulsifying Agent Solutions

46. Solutions of Span 20, Span 40, Span 60 and Span 80 in Perclene were stored at 5°C., room temperature and 45°C. for 4 weeks. The results are given in Appendix A, Table XVI. At 5°C there was considerable separation of Span 40 and Span 60 from a 5% solution. The solutions of Span 20 and Span 80 showed no separation except for a slight amount of insoluble material which did not increase in amount on standing. Solutions filtered before storage did not show this slight separation.

47. Perclene solutions of Span 20 (5%) and two different samples of Span 80 (each 10%) were placed at -20°F for 24 hours. All the samples were frozen. When warmed to room temperature each melted with the separation of the Spans which redissolved easily when the mixtures were agitated. The samples were again stored at -20°F. After 48 hours the temperature was raised to 0°F. The solutions were homogeneous when examined 24 hours after raising the temperature. The samples were then stored at room temperature for 5 months. There was no separation or evidence of reaction or decomposition of the components.

48. Solutions of Span 20 and Span 80 in Perclene made satisfactory emulsion pastes after eight months storage at room temperature, which included summer temperatures exceeding 30°C at times. There was no evidence that any deterioration of the solutions had occurred.

(2) Storage of Prepared Emulsion Pastes

49. The stabilities of emulsion pastes made with Span 20 or Span 80 and S-461, S-210 or RH-195, stored at room temperature and 45°C are tabulated in Appendix A, Table XVII. Examination of the data shows that S-210 pastes lost no active chlorine when kept at room temperature for 8 weeks. S-461 pastes made with Span 20 were likewise stable for 8 weeks but those made with Span 80 began to deteriorate after 4 weeks at room temperature. Pastes made with RH-195 were somewhat less stable than those made with S-461.

50. At 45°C, the S-210 pastes did not show loss of active chlorine even after 8 weeks. S-461 pastes were fairly stable for two weeks and RH-195 pastes showed considerable loss of active chlorine after one week.

(3) Addition of Thickening Agents

51. A number of formulations for emulsion pastes were made using polybutene and aluminum soaps as thickeners. The thickeners were dissolved in the Perclene after solution of the Span. In every case the resulting emulsion was easy to prepare, smooth and gave more effective adhesion to painted panels. Decontamination of H was effective, but the dried paste films containing the thickeners were more difficult to remove completely from the panels, except where aluminum oleate was the thickener. Because of the better adhesion, it is to be recommended that, if emulsion pastes are considered for use, a thickener such as aluminum oleate be added to the extent of 1 or 2% of the Perclene solution. Such solutions including Span 20 or Span 80 show no separation or deterioration after storage for 8 months at 5°C.

(4) Corrosive Action of Emulsion Pastes on Metals

52. It was observed that the emulsion pastes were less corrosive to metals than was TCE/RH-195 solution. The NDRC group studying decontamination at the DuPont Experimental Station also observed this and stated that the emulsion pastes were about as corrosive as the Perclene/potassium oleate/S-461 paste. This latter system was found to be much less corrosive than TCE/RH-195 solution.

(5) Storage of S-461 in Perclene/Span 80 Solutions

53. S-461 powder is believed to be unsuitable for storage aboard ship because of its inflammability. Numerous attempts have been made at this and other laboratories to reduce the hazard by the addition of water, hydrated salts and inert materials such as bentonite. None of these attempts proved successful from a practical standpoint. Likewise mixtures of S-461 with as much as 60% of a non-inflammable chloroamide, such as S-210 and S-436 did not inhibit the decomposition of S-461 sufficiently.

54. Because of the stability of the emulsion pastes to loss of active chlorine, it seemed possible that S-461 might be stored in the Perclene/Span solution. Accordingly, mixtures of S-461 in Perclene/Span 80 solution were put in storage at room temperature, 45°C and 60°C. The storage tests were dropped after 3 weeks because of lack of stability of the mixtures due to loss of active chlorine. The results are given in Appendix A, Table XVIII.

(6) Attempts to Camouflage Emulsion Pastes

(a) Addition of Dispersible Pigments

55. A number of S-461 emulsions were prepared using Span 80 as the emulsifying agent and with a Perclene/water ratio of 20/30, 20/50 or 20/80. DuPont numbers TLX-1A and TLX-68 were blue pigment pastes and TLX-69 was a carbon black paste. Suitable blue-gray pastes could be prepared by mixing one of the blue pigments with TLX-69. The addition of TLX-1A made it difficult to prepare the emulsions and the resulting emulsions were less stable. TLX-68 and TLX-69 did not affect the emulsions. It has been reported by the NDRC that carbon black cannot be stored with chloroamides without loss of active chlorine. Consequently, attempts were made to store TLX-68 and TLX-69 as a dispersion in the Perclene/Span solution. TLX-68 alone showed very little separation after three weeks, but TLX-69 and a mixture of TLX-68 and TLX-69 showed considerable separation within a few days. Because of this separation, other means of camouflaging the pastes were sought.

(b) Addition of Perclene-soluble dyes

56. A number of oil soluble black and blue dyes were obtained and tested as follows. One gram of the dye was dissolved in 100 g. of Perclene. Solubility and color of the solution was noted. Emulsion pastes were then prepared using Span 80, S-461 and a 20/50 ratio of Perclene/water. The color of the paste, both wet and dry, was recorded. The results are shown in Appendix A, Table XIX. Mixtures of Oil Black 5115 PDR, Oil Black 24087 or Calco Oil Black 8603 with AZP Oil Blue Black B, Calco Oil Blue Goo or Sudan Blue GA 247459 incorporated in the pastes gave satisfactory camouflage colors. Storage of these dyes in Perclene/Span 80 solutions was satisfactory. It was concluded that the use of Perclene soluble dyes was the most satisfactory way of coloring the pastes.

PART II. COMPARISON OF THE EMULSION PASTES WITH OTHER DECONTAMINATING SYSTEMS

A. Decontamination Efficiency

57. A number of experiments were carried out in an attempt to determine, for each of four decontamination systems, the relative coverage in sq.yd./gal. which would effectively decontaminate H in deck paint. At the same time observations were made regarding the ease of handling the various systems. The systems used were TCE/RH-195 solution, S-461 or S-210 emulsion pastes, Perclene/potassium oleate/S-461 paste and a modified potassium oleate paste. This improved potassium oleate paste is described in NDRC Report OSRD No. 3927, "Improved Decontaminating Systems", August 1, 1944. The NDRC (DuPont Group) found that the addition of a special wax (Aristowax 160/165) to the potassium oleate/S-210 paste considerably improved its decontaminating efficiency and made it somewhat easier to remove than earlier pastes of this type.

58. Each of these systems would require the storage of two components. Representative formulations used were as follows:

System	Parts by Weight	
	Component A	Component B
TCE/RH-195	1 part RH-195	10 parts TCE
Emulsion Paste *	1 part S-461	2.6 parts Perclene 0.3 parts Span
K Oleate Paste	1 part S-461	3.3 parts Perclene 1.7 parts K Oleate
Improved K Oleate Paste	1 part S-210	6.53 parts Perclene 1.75 parts K Oleate (28% H_2O) 0.27 parts $Ba(OH_2).8H_2O$ 0.15 parts Aristowax 160/165

* Exclusive of water used for make-up.

59. Navy 20B blue deck paint on steel was the surface used in all the decontamination tests. Small panels (2" x 2" and 6" x 12") were used indoors at room temperature and outdoors at temperatures as low as 45°F. A 3-inch Navy gun and mount was decontaminated at temperatures from 45°F to 80°F. Steel panels, 36" x 72", were used in experiments, most of which were carried out in a large gas chamber at 95°F. In all cases where used, TCE/RH-195 was sprayed on the contaminated panels. In the larger scale tests both the Navy 3-gallon and the Army 3-gallon sprayer were used. The Army sprayer was the more convenient to use in applying the solution efficiently to a small area. In the larger scale tests, the pastes were sprayed with the 3-gallon sprayers or applied with a paint brush. Details of some of these experiments are given in Appendix B.

60. On the basis of the data from Appendix B, it was estimated that, with most efficient usage of each system, H contamination of 0.75 oz/sq.yd. could be decontaminated safely by the application of 0.7 lb. of TCE/RH-195 solution, 1.5 lb. of S-461 emulsion paste, 1.5 lb. of K Oleate/S-461 paste, or 0.9 lb. of Improved K Oleate paste per square yard of contaminated surface. The pastes were applied by brushing in most cases. The TCE/RH-195 was sprayed twice with a 10-minute interval between sprayings. The density of each system was found to be: TCE/RH-195, 13.2 lb/gal.; emulsion paste, 9.5 lb./gal; K Oleate/S-461 paste, 11.3 lb/gal; and Improved K Oleate paste, 12.1 lb./gal. From these figures was calculated the coverage in sq.yd./gal. for each system. These figures are tabulated below:

System	Lb/yd^2 Needed	Density Lb./Gal.	Coverage Sq.Yd/gal.	Coverage Sq.yd./3 gal. sprayer
TCE/RH-195	0.7	13.2	19.0	57
Emulsion Paste	1.5	9.5	6.3	19
K Oleate Paste	1.5	11.3	7.5	22
Improved K Oleate	0.9	12.1	13.4	40

From these figures, it is evident that TCE/RH-195 solution would be the most convenient to use because it will cover a larger area than that covered by a similar volume of paste.

B. Storage Requirements

61. Using the values for the coverage by each system obtained in the preceding paragraph, the relative weight and volume requirements necessary for decontaminating a square yard of surface would be as follows:

	Weight to be carried (Pounds)		
	Solvent or Solvent Mixture	Chloroamide	Total
Emulsion Paste, S-461	0.39 *	0.14	0.53
TCE/RH-195	0.63	0.07	0.70
K Oleate Paste, S-461	1.25	0.25	1.50
Improved K Oleate Paste	0.81	0.09	0.90

* This does not include 0.97 lb. of water needed.

	Relative Volume to be carried		
	Solvent or Solvent Mixture	Chloroamide**	Total
Emulsion Paste, S-461	0.24	0.34	0.58
TCE/RH-195	0.39	0.14	0.53
K Oleate Paste, S-461	1.00	0.62	1.62
Improved K O paste, S-210	0.56	0.18	0.74

** Assuming density of S-210 and RH-195 is 0.5 and of S-461 is 0.4.

62. These calculations show that the paste systems would require as much or more space to store the necessary ingredients than is the case with TCE/RH-195. Furthermore, if S-210 were used in the emulsion paste somewhat more paste would be required per unit area. Also it should be pointed out that for a lighter contamination of H, it might be possible to apply the TCE/RH-195 more thinly. It is doubtful that the emulsion paste could be effective in a much thinner layer because the paste would dry too quickly. The improved potassium oleate paste could be used in a somewhat thinner layer because the special wax retards the evaporation of Perclene.

C. Preparation and Application

63. The TCE/RH-195 solution is readily mixed and easily applied by spraying or with swabs. The prepared solution is stable for several weeks at moderate temperatures, but should ordinarily be prepared shortly before use. The residues should be hosed off with water, but this offers no difficulty as the residues are readily washed away.

64. The emulsion paste is readily mixed and may be applied by spraying or brushing. It is more apt to clog any spray nozzle than is the TCE/RH-195 solution, and it is more difficult to clean from the 3-gallon sprayers. The prepared paste is stable for several months at moderate temperatures, but it is not recommended that it be prepared in advance. The residues are usually readily dislodged by hosing with water but may require scrubbing with a brush in some cases. The residues are much more voluminous than for TCE/RH-195, and disposal of them will require more attention to drainage facilities.

65. The Perclene/potassium oleate/S-210 paste is easily mixed but requires more stirring than do the other two systems. The solvent mixture, before adding the S-210, will separate to some extent on storage and is difficult to remix when cold (35° - 40°F.). The prepared paste will lose much of its active chlorine within twenty-four hours and must be prepared fresh shortly before use. This paste also is more apt to clog the spray nozzle and is difficult to remove from the 3-gallon sprayer. The residues are more difficultly removed than for the other two systems and require scrubbing in addition to hosing with water. The residues from this paste are likewise voluminous.

D. Other Considerations

66. Both the emulsion paste and potassium oleate paste would require the addition of dyes or pigments to camouflage the white color of the pastes or paste residues. Satisfactory means have been worked out for doing this. TCE-RH-195 solution requires no camouflage. Also, caking of the chloroamide would increase somewhat the work of stirring up the pastes, as compared to the TCE-RH-195 solution.

E. Summary of Comparison of Systems

67. The comparisons of the three best systems under discussion are summarized in Table XX. From this table it is seen that the chief disadvantages of the TCE/RH-195 solution are its toxicity, corrosive effects on metals and harmful effect on plastics. The pastes are much better in these respects but suffer from

lower covering power, somewhat greater storage requirements and greater difficulty in application and removal. It should be noted also that S-210, which is used in the pastes, is not in production at present although it can be manufactured on a large scale.

68. The potassium oleate paste is a better decontaminant for the nitrogen mustards than are the other two systems. The potassium oleate paste will decontaminate CN, for which the other two systems are ineffective.

TABLE XX

Comparison of Decontamination Systems for Use on Shipboard

	TCE/RH-195 Solution	Improved K Oleate/S-210	Emulsion/S-210 Paste
Decontamination of H	Good	Good	Good
Decontamination of L	Good	Good	Good
Decontamination of HN	Fair	Fair	Poor
Decontamination of CN	Poor	Good	Poor
Decontamination of BBC	Poor	Fair	Poor
Decontamination of wet surfaces	Poor	Good	Good
Decon. at high temp. (95°F)	Good	Good	Good
Decon. at low temp. (50°F.)	Good	Good	Good
Inflammability of Materials	None	None	None
Availability of Materials	Good	Good *	Good *
Lowest usable temperature	About -30°F	About -4°F	About 32°F
Camouflage necessary	No	Yes	Yes
Toxicity of vapors	Bad	Good	Good
Skin irritation	Bad	Bad, if hot	Bad, if hot
Ease of Preparation	Good	Fair	Fairly good
Stability of Mixture	Good	Fair	Good
Ease of Application	Good	Fair	Fair
Ease of Removal	Good	Requires scrubbing	May need scrubbing
Disposal of Residues	Good	Bad	Fair
Effect on deck paint	Bad	Little effect	Little effect
Corrosion	Fair	Good	Good.
Effect on plastics	Bad	No effect	No effect
Relative vol. of materials required	1.0	1.4	1.1
Covering power, yd^2/3-gal. sprayer	57	40	19

* S-210 is not in production at present date.

SUMMARY AND CONCLUSIONS

69. A new decontamination system, the emulsion paste, has been developed which consists essentially of a dispersion of a chloroamide in a viscous emulsion of water in Perclene. Water makes up 65% of the total weight of the system. The chloroamide giving best practical results is S-210. With the exception of its inflammability, S-461 is better than S-210. Span 20 (sorbitan monolaurate) and Span 80 (sorbitan monooleate) were found to be the best emulsifying agents for this purpose.

70. This system will satisfactorily decontaminate H in deck paint under various conditions. It will decontaminate L and HV readily, and will partially decontaminate HN-3 and HN-1, being as effective as TCE/RH-195 in this respect.

71. Cellulose acetate and methyl methacrylate clear plastics are readily decontaminated from H by the emulsion paste without further damage than that caused by H. Aircraft paints are fairly readily decontaminated from H, although M485 lacquer is softened and loosened by the paste. Cellulose nitrate doped fabric is decontaminated and cellulose acetobutyrate doped fabric partially decontaminated without damage by the paste to the coating or fabric.

72. The emulsion paste system is much less toxic, less corrosive to metals and less harmful to paint and clear plastics than is TCE/RH-195. The emulsion paste is not quite as easy to handle as TCE/RH-195 and would require slightly more storage space. The mixed paste has about one-third as much effective covering power per gallon as does the TCE/RH-195 solution.

73. An improved modification of the NDRC potassium oleate paste has been examined and found to be about twice as efficient as the earlier potassium oleate paste. This improved paste would require more storage space than either the TCE/RH-195 solution or the emulsion paste, and is more difficult to remove after decontamination. The improved potassium oleate paste has about twice as much covering power per gallon as does the emulsion paste.

RECOMMENDATIONS

74. None. The TCE/RH-195 solution is more efficient and is easier to handle than is any paste system thus far examined. The paste systems warrant consideration if the toxicity and deleterious effect of the TCE/RH-195 solution on many surfaces should ever make it necessary to use some other system.

APPENDIX A

TABLE I

Manufacturer	Agent	State	Solubility of 1 gm. Agent in 10 gm. Perclene	Ease of Emulsification	Emulsion Type	Stability of Emulsion	Thickening Action
Amer. Cyan.	Aerosol OT	solid	completely, slowly	easy	O/W	-	none
DuPont	Alkanol SA	solid	almost completely	easy	O/W	bad	none
-	Aluminum mahogany sulfonate	liquid	completely	no emulsion	-	-	-
-	Aluminum naphthenate	solid	completely on heating	" "	-	-	-
-	Aluminum oleate	liquid	completely	" "	-	-	-
-	Alum. palmitate	solid	completely on heating	"	-	-	-
-	Alum. sebacate	solid	"	" "	-	-	-
-	Alum. stearate	solid	"	" "	-	-	-
-	Aresklene 375	liquid	completely, slowly	easy	O/W	bad	none
Atlas	Arlacel A	liquid	completely, quickly	"	W/O	good	good
"	Arlacel C	liquid	completely, quickly	"	W/O	good	good
"	Arlex	solid	insoluble	-	-	-	-
DuPont	Avitex SF	soft paste	disperses readily	easy	O/W	bad	none
	Barium oleate	solid	partially	no emulsion	-	-	-
Ohio Apex	Butoxyethyl stearate	liquid	completely, quickly	difficult	-	bad	none
Kessler	Butyl laurate	liquid	completely, quickly	difficult	-	bad	none
Kessler	Butyl stearate	liquid	completely, quickly	difficult	-	bad	none
	Calcium oleate	solid	partially	no emulsion	-	-	-
DuPont	Chlorinated alkyl sulfonic acid	liquid	partially	easy	-	bad	none
	Cobalt naphthenate	solid	slowly when heated	no emulsion	-	-	-
	Copper oleate	solid	slowly when heated	difficulty	-	bad	none

TABLE I (Cont'd.)

Manufacturer	Agent	State	Solubility of 1 gm. Agent in 10 gm. Perclene	Ease of Emulsification	Emulsion Type	Stability of Emulsion	Thickening Action
Vanderbilt	Darvan No. 1	solid	insoluble	-	-	-	-
"	Darvan No. 2	"	"	-	-	-	-
Dewey & Almy	Daxad No. 23	"	"	-	-	-	-
Glyco	Diethyleneglycolde-caricinoleate	liquid	complete	fair	W/O	fair	good
"	Diethyleneglycol monoethyl ether laurate	paste	completely	difficult	-	bad	none
"	Diethyleneglycol monoethyl ether stearate	solid	completely	difficult	-	bad	none
"	Diglycol laurate	liquid	completely	fair	W/O	fair	some
"	Diglycol ricinoleate		completely, cloudy	difficult	W/O	bad	-
"	Diglycol stearate	solid	partially	fair	W/O	bad	slight
Gen.Dyestuff	Emulphor A	liquid	completely	easy	O/W	fair	slight
" "	Emulphor AG oil soluble	"	completely, cloudy	"	O/W	bad	slight
" "	Emulphor ELA	"	" "	"	O/W	-	none
" "	Emulphor ON	solid	" "	"	O/W	fair	slight
Glyco	Ethyleneglycol monoethyl ether laurate	liquid	completely	difficult	-	bad	none
	Gelatin	solid	insoluble	none	-	-	-
Glyco	Glyceryl oleyl triricinoleate	liquid	complete	fair	W/O	-	good
Kessler	Glyceryl monolaurate	solid	partially	"	W/O	good	"
Glyco	Glyceryl monoricinoleate	"	completely, turbid	easy	W/O	"	"
Kessler	Glyceryl monostearate	"	partial	fair	W/O	"	"
Glyco	Glyceryl trilevulinyl triricinoleate	liquid	complete	difficult	W/O	poor	-

TABLE I (Cont'd.)

Manufacturer	Agent	State	Solubility of 1 gm. Agent in 10 gm. Perclene	Ease of Emulsification	Emulsion Type	Stability of Emulsion	Thickening Action
Eimer & Amend	Gum tragacanth	solid	insoluble	no emulsion	-	-	-
	Hercose C High	"	"	" "	-	-	-
Horn	Hornkem No. 3	liquid	"	" "	-	-	-
Rohm & Haas	Hydrocide 10-X	"	none or slight	easy	O/W	bad	none
Gen.Dyestuff	Igepal CA	"	" " "	"	"	-	"
" "	Igepon T gel	gel	partial	"	"	bad	"
	Imerial TX	paste	"	"	"	"	slight
Mallinkrodt	Lanolin	"	complete but slow	fair	W/O	"	none
	Lead oleate	"	" if heated	no emulsion	-	-	-
	" stearate	solid	" " "	" "	-	-	-
	" resinate	"	partial	fair	W/O	good	some
	" ricinoleate	paste	complete, if heated	no emulsion	-	-	-
	Magnesium oleate	solid	complete, if heated	no emulsion	-	-	-
	Manganese naphthenate	"	complete	" "	-	-	-
Atlas	Mannitan diricinoleate	liquid	complete	easy	W/O	good	good
Atlas	Mannitan monolaurate	liquid	complete	easy	W/O	good	good
Atlas	Mannitan monooleate	liquid	complete	easy	W/O	good	good
Atlas	Mannitan tetraricinoleate	"	complete	easy	W/O	good	good
Atlas	Mannitan triricinoleate	"	complete	easy	W/O	good	good
Atlas	Modified sorbitan monooleate	liquid	complete, turbid	easy	O/W	fair	none

TABLE I (Cont'd)

Manufacturer	Agent	State	Solubility of 1 gm. Agent in 10 gm. Perclene	Ease of Emulsification	Emulsion Type	Stability of Emulsion	Thickening Action
Allied	Nacconol EP	solid	partial	easy	O/W	bad	none
Allied	" NR	"	"	"	"	"	"
Glyco	Monoethylene-glycol diricinoleate	liquid	complete	"	"	-	"
Wolf	Orotol N	solid	partial	"	"	-	"
Proctor & Gamble	Orvus	paste	insoluble	"	"	bad	"
Parke,Davis	Phemerol	solid	partial	"	"	-	"
A.D. Mackay	Potassium laurate	paste	"	"	-	bad	"
DuPont	" oleate	paste	"	"	-	"	"
	" salts of linseed oil acids	solution	"	"	-	"	"
Beacon	Potassium stearate	paste	"	"	W/O	"	"
	Ramapol WF	liquid	complete	difficult	"	"	some
Gen.Dyestuff	Ramasol NA	solid	insoluble	no emulsion	-	-	-
	Rosin	"	partial	difficult	-	poor	-
Monsanto	Santomerse #3	"	complete	easy	O/W	"	slight
"	Santomerse S	liquid	none to slight	"	-	"	"
Sherwood Ref'g. Co.	Sherosope L	"	complete	difficult	-	"	"
" "	Sherosope N	"	"	"	-	"	"
"	Sherosope T	"	"	"	-	"	"
Ciba Co.	Solvadine NC conc.	"	none to slight	easy	O/W	"	none
Atlas	Sorbide dipalmitate	solid	complete	fair	W/O	fair	some
Atlas	Sorbide monolaurate	liquid	"	easy	"	good	good
Atlas	Sorbide monostearate	solid	"	"	"	"	"
Atlas	Sorbitan monolaurate (Span 20)	liquid	"	"	"	"	"

TABLE I (Cont'd)

Manufacturer	Agent	State	Solubility of 1 gm. Agent in 10 gm. Perclene	Ease of Emulsification	Emulsion Type	Stability of Emulsion	Thickening Action
Atlas	Sorbitan monooleate (Span 80)	liquid	complete	easy	W/O	good	good
Atlas	Sorbitan monopalmitate (Span 40)	solid	"	"	"	"	"
Atlas	Sorbitan monoricinoleate	liquid	"	"	"	"	"
Atlas	Sorbitan monostearate (Span 60)	solid	"	"	"	"	"
Atlas	Sorbitan trilaurate	liquid	"	difficult	"	"	slight
Atlas	Sorbitan trioleate	"	"	fair	"	fair	some
Atlas	" tetralaurate	"	"	difficult	-	poor	-
Glyco	Sorbitol laurate	"	"	easy	W/O	fair	good
Glyco	Soya lecithin	"	"	difficult	-	poor	slight
Glyco	Tetraethyleneglycol monostearate	solid	"	easy	O/W	-	none
Glyco	Tetraethyleneglycol distearate	"	"	fair	W/O	poor	slight
Rohm & Haas	Triton B 1956	liquid	"	easy	W/O	good	good
Rohm & Haas	" H-2994	solid	"	"	O/W	fair	none
Rohm & Haas	" K-60	paste	partially	difficult		poor	"
Rohm & Haas	" NE	liquid	complete, turbid	easy	O/W	-	none
Eimer & Amend	Turkey red oil	"	insoluble	"	"	-	"
Atlas	Tween 60	"	partial	"	"	-	"
Gen.Dyestuff	Vultramine	solid	insoluble	no emulsion	-	-	-
	Zinc oleate	"	partial	difficult	-	-	-

TABLE II

Classification of Emulsifying Agents

These agents were arbitrarily divided on the basis of ease of emulsification of water in Perclene by hand stirring, and their stability in the presence of S-461.

Most Promising

Arlacel A

Arlacel B

Mannitan diricinoleate

Mannitan monolaurate

Mannitan triricinoleate

Sorbitan monooleate (Span 80)

Sorbitan monopalmitate (Span 40)

Sorbitan monostearate (Span 60)

Intermediate

Diglycol laurate

Glyceryl monoricinoleate

Mannitan monooleate

Mannitan tetraricinoleate

Sorbide monolaurate

Sorbide monostearate

Sorbitan monolaurate (Span 20)

Sorbitan monoricinoleate

Sorbitol laurate

Triton B1956

Least Promising

Glyceryl monolaurate

Glyceryl monostearate

Glyceryl oleyl triricinoleate

Lead resinate

Sorbide dipalmitate

Sorbitan trioleate

TABLE III

Decontamination of H in Deck Paint

The ratios 20/30, 20/50 and 20/80 refer to the Perclene/water ratio. Tests were repeated 24 hours later if positive one hour after decontamination.

Emulsifying Agent	Paper Test Times (Minutes)									
	20/30 S-461		20/30 RH-195		20/30 S-210		20/50 S-461		20/80 S-461	
	1 Hr.	24 Hrs.	1 Hr.	24 Hrs.	1 Hr.	24 Hrs.	1 Hr.	24 Hrs.	1 Hr.	24 Hrs.
Arlacel A	neg.		neg.		24 neg.	neg.	6 neg.	neg.	-	-
Arlacel C	neg.		neg.		neg.		neg.		neg.	
Mannitan Diricin-oleate	neg.		2 1	neg. neg.	neg.		12 neg.	neg.	-	-
Mannitan Monolaurate	neg.		neg.		neg.		neg.		-	-
Mannitan Triricin-oleate	15 1	neg. neg.	2 1	neg. neg.	neg.		1 5	neg. neg.	-	-
Span 80	neg.		22 37	neg. neg.	19 neg	neg.	1 neg.	neg.	neg.	
Span 40	neg.		neg.		24 3	neg. neg.	neg.		neg.	
Span 60	neg.		neg.		2	neg.	neg.		neg.	
Typical Blank	<1 <1	6 5								

TABLE IV

Decontamination of H in Deck Paint with 20/80 Emulsion Pastes of S-210

Emulsifying Agent	% of Weight of Perclene	Paper Test Times 1 Hr.	24 Hrs.
Span 40	1%	neg.	
	5%	2	neg.
		2	neg.
	10%	2	neg.
		5	neg.
Span 80	1%	2	neg.
		3	neg.
	10%	2	neg.
		neg.	

TABLE V

Decontamination of H in Deck Paint

Perclene/Water Ratio	Percentage of Chloroamide	Paper Test Times With Span 20 Immed.	With Span 20 24 Hrs.	With Span 80 Immed.	With Span 80 24 Hrs.
20/50	5% S-330	18	neg.	neg.	
		10	neg.	neg.	
20/50	10% S-330	1	6	neg.	
		1	9	neg.	
20/50	5% S-222	1	2	1	30
		1	2	1	9
20/50	10% S-222	1	3	1	3
		1	neg.	1	3
20/50	10% S-436	1	9	1	11
		1	8	1	neg.
20/30	10% S-330	neg.		neg.	
		6	neg.	neg.	
20/30	10% S-222	1	neg.	3	neg.
		1	12	3	neg.

TABLE VI

Decontamination of H in Deck Paint

Emulsion Paste	Application Time (Min.)	Test Time Immed.	3 Hrs.	24 Hrs.
Span 20/S-210	20	1 1	5 2	neg. 15
Span 20/S-210	40	9 30	neg. neg.	
Span 20/S-210	60	neg. neg.		
Span 80/S-210	20	1 1	5 10	neg. neg.
Span 80/S-210	40	neg. neg.		
Span 80/S-210	60	neg. neg.		

TABLE VII

Decontamination of H on Deck Paint

Emulsion Paste	Paper Test Time (Min.) Immed.	24 Hrs.
* HCB/Span 20/S-461	<1 <1	2 4
HCB/Span 20/RH-195	<1 <1	1 1
HCB/Span 20/S-436	1 3	8 neg.
* HCP/Span 20/S-436	3 5	21 30
* CP/Span 20/RH-195	<1 <1	
CP/Span 20/S-461	<1 <1	
CP/Span 20/S-436	<1 <1	
* BL/Span 20/S-461	<1 <1	

*HCB = Hexachlorobutadiene
HCP = Hexachloropropene
CP = Chloropropane
BL = Butyl laurate

TABLE VIII

Solubility of Chloroamides in Hexachlorobutadiene and Hexachloropropane

Solvent	Chloroamide	Solubility GM./100 cc. Sol'n.
Hexachlorobutadiene	S-461	0.015
"	RH-195	0.82
"	S-436	4.85
Hexachloropropene	S-461	0.03
"	RH-195	1.00
"	S-436	5.40

TABLE IX

Decontamination of H in Wet Deck Paint Panels

The pastes were prepared with S-461, Perclene, Water and the emulsifying agents listed.

Emulsifying Agent	Paper Test Times (Minutes) Wet with Fresh Water	Wet with Sea Water
Arlacel C	neg.	neg.
Span 80	neg.	neg.
Span 40	neg.	neg.
Span 60	neg.	neg.
Mannitan monolaurate	-	neg.

TABLE X

Decontamination of H in M485 Airplane Lacquer

Emulsifying Agent	Application Time (Min.)	Paper Test Time (Minutes) S-210 Immed.	S-210 2 Hrs.	S-461 Immed.	S-461 2 Hrs.
Span 20	20	2	28	1	3
		4	42	1	1
Span 80	20	6	neg.	1	4
		4	42	2	neg.
Span 20	60	2	8	neg.	
		6	neg.	neg.	
Span 80	60	2			
		neg.			

TABLE XI

Decontamination of H in Doped Fabrics

Emulsion Paste	Application Time (Hrs.)	CN *Fabric Immed.	CAB *Fabric Immed.	CAB *Fabric 24 Hrs.
Span 20/S-210	1	5	1	4
	2	-	1	9
Span 20/S-461	1	neg.	1	4
	2	-	1	neg.
Span 80/S-210	1	neg.	2	5
	2	-	1	6
Span 20/S-461	1	neg.	1	5
	2	-	1	29

*CN = cellulose nitrate doped
*CAB= cellulose acetate butyrate doped

TABLE XII

Tensile Strength Measurement of Doped Fabrics after 3 Successive Treatments and 2 Weeks' Outdoor Weathering

Treatment	Tensile Strength (Lbs.) Cellulose Nitrate	Cellulose Acetobutyrate
1. Blank	113	127
2. Span 20/S-210 Emulsion	107	120
3. Span 20/S-461 Emulsion	115	131
4. Span 80/S-210 Emulsion	113	117
5. Span 80/S-461 Emulsion	106	120
6. H + (2)	116	130
7. H + (3)	113	129
8. H + (4)	110	129
9. H + (5)	114	125

TABLE XIII

Decontamination of HV and L on Deck Paint with S-461 20/30 Perclene/water Pastes

Emulsifying Agent	Paper Test Time (Min.) HV Immed.	HV 24 Hrs.	L Immed.
Arlacel C	23 15	neg. neg.	neg.
Mannitan monolaurate	neg.		neg.
Span 80	neg.		neg.
Span 40	neg.		neg.
Span 60	neg.		neg.

TABLE XIV

Decontamination of HN-3

Decontamination System	Paper Test Times (Min.) Immed.	24 Hrs.	48 Hrs.
Span 80/S-461 emulsion paste	10		24
	10		27
Span 80/S-210 emulsion paste	10		21
	10		21
Spray twice, TCE/RH-195	10		24
	10		27
Spray twice, TCE/S-436	26	neg.	
	26	neg.	
Blank	3	10	20
	3	10	19

TABLE XV

Decontamination of HN-1

Decontamination System	Paper Test Times (Min.) Immed.	3 Hrs.	48 Hrs.
Span 80/S-461 emulsion paste	3	6	16
	4	6	16
Span 80/S-210 emulsion paste	5	6	16
	5	6	17
Span 20/S-436 emulsion paste	neg.		
	neg.		
Span 80 emulsion, no chloroamide	4	5	16
	5	6	20
Spray twice, TCE/RH-195	4	4	11
	3	4	9
Spray twice, TCE/S-436	5	23	neg.
	6	23	neg.
Blank	2	1	5
	2	1	5

TABLE XVI

Storage of Perclene/Span Solutions

Perclene Solutions	% Span	Separation on Storage (4 Wks.) R.T.	5°C	45°C
Span 20	10	None	V. slight	None
Span 20	5	None	None	None
Span 40	5	V.slight*	Considerable	V. slight
Span 60	5	V.slight	Considerable	V. slight
Span 80	5	V.slight	V. slight	V. slight

*V. slight separation appears to be due to insoluble impurities such as sorbitol.

TABLE XVII

Stability Of Emulsion Pastes on Storage

Storage Temp.	Emulsifying Agent	Chloroamide	% Cl^+ Orig.	1 Wk.	2 Wks.	4 Wks.	8 Wks.
R.T.	Span 20	S-461	2.80	2.79	2.85	2.83	2.82
	" 20	S-210	1.85	1.86	1.87	1.90	1.87
	" 20	RH-195	3.01	3.03	2.98	2.96	2.62
	Span 80	S-461	2.56	2.62	2.62	2.54	2.41
	" 80	S-210	1.78	1.78	1.79	1.78	1.80
	" 80	RH-195	1.85	1.82	1.79	1.74	1.57
45°C	Span 20	S-461	2.80	2.74	2.62	1.91	0.28
	" 20	S-210	1.85	1.91	1.94	-	1.67
	" 20	RH-195	3.01	2.30	1.49	-	-
	Span 80	S-461	2.56	2.44	2.41	2.31	1.53
	" 80	S-210	1.78	1.77	1.78	-	1.62
	" 80	RH-195	1.85	1.34	0.87	-	-

TABLE XVIII

Storage of S-461/Span 80/Perclene Mixtures

Storage Time	% Cl^+ R.T.	45°C	60°C
Original	16.52	16.37	16.30
3 days	15.62	15.20	14.80
7 days	16.10	15.58	14.34
22 days	16.14	15.00	13.43

TABLE XIX

Tests of Oil Soluble Dyes in Emulsion Pastes

Manufacturer	Dyestuff	Solubility in Perclene	Color in Perclene	Color of Paste on Panel, Wet	Color of Paste on Panel, Dry
National Aniline	Oil Black 38226 PDR	soluble	Bluish-black	Chocolate	Chocolate
" "	AZP Oil Blue Black B	Soluble	Red-black	Dark purple	Dark blue-purple
" "	Nubian Resin Black	Sl.soluble	Brown-purple	Pink	Cream
" "	Oil Black 5115 PDR	Soluble	Purple-black	Brown	Dark brown
" "	Alizarine Blue	Soluble	Dark blue	Light blue	Light blue
" "	Spirit Nigrosine R	Insoluble	-	-	-
" "	Oil Black 24087	Soluble	Dark blue	Brown	Dark brown
Gen. Dyestuff	Induline Base R	Soluble	Dark red	Light brown	Light brown
" "	Bismarck Brown TSS Base, 35356	Sl.soluble	Red-brown	-	-
" "	Sudan Blue GA, 247459	Soluble	Blue black	Blue	Blue
" "	Induline Base B 24002	Sl.soluble	-	-	-
" "	Alizarine Sky Blue B, 34272	Soluble	Dark blue	Blue	Light blue
Calco	Calcofast Black Green Toner	Insoluble	-		
"	Calco Oil Black B-3804	Soluble	Gray	Brown	Light brown
"	Calco Nigrosine Base	Soluble	Red-black	Pink	Pink
"	Victoria Blue B Base	Soluble	Red-black	Dark blue	Light blue
"	Calcofast Spirit Black R	Insoluble	-	-	-
"	Calco Oil Black 8603	Soluble	Purple-black	Blue-gray	Blue-gray
"	Calco Oil Blue 10641	Soluble	Dark bluegreen	Blue-green	Light blue
"	Calco Oil Blue IR Base	Soluble	Purple	Purple	Violet
"	Calco Oil Blue NA	Soluble	Blue	Dark blue	Blue
"	Calco Oil Blue Goo	Soluble	Blue green	Light blue	Light blue
"	Calcochrome Alizarine Blue AWR	Sl.soluble	-	-	-
"	Calco Oil Blue B5199	Soluble	Blue-black	Steel gray	Steel gray
"	Calco Oil Black F4160	Soluble	Blue-black	Brown	Light brown

APPENDIX B

Decontamination Efficiency of TCE/RH-195 and Paste Systems

1. A series of decontamination experiments was made with the paste systems using 6" x 12" blue deck painted steel panels. Each panel was contaminated on 2-inch squares with H at a density of about 20 g./yd^2. The temperature during the tests was 95°F. One hour after the H was applied, the pastes were brushed on, left for one hour then removed by hosing with water and brushing if necessary. The results of the tests for residual H are given in Table I.

TABLE I

Efficiency of Decontaminating Systems (Small Panel)

Paste	Weight of Paste Used Lb./yd^2	Paper Test Time (Minutes) Immed.	3 Hrs.	24 Hrs.
E.P. S-461	1.6	neg.	-	-
" "	1.1	8	-	neg.
" "	0.7	1	-	neg.
E.P. S-210	1.7	5, neg.	neg.	-
" "	1.1	2. 2	14, 28	neg.
" "	0.8	<1	4	neg.
K Oleate/S-461	0.9	4	-	neg.
" "	0.6	2	-	neg.
" "	1.2	11	-	neg.

2. A series of experiments using several decontaminating systems were run in which 36" x 72" blue deck painted panels were used. Data were obtained for TCE/RH-195, the potassium oleate paste and emulsion pastes. The H contamination was about 20 g./yd^2. The temperature during the tests was 85 to 95°F. The tests for residual H are given in Table II.

TABLE II

Efficiency of Decontaminating Systems (Large Panels)

System Used	Application Method	Weight Used Lb/yd	Paper Test Time (Min.) Immed.	3 Hrs.
E.P. Span 80/S-210	Spray	2.7	<1(8)	2 to neg.
E.P. Span 80/S-210	Brush	0.5	<1(4)	1,<1(3)
E.P. Span 80/S-461	Brush	0.4	3,1,1,2	5,3,2,3
E.P. Span 80/S-461	Brush	1.25	2,4,6,neg.	-
K. Oleate/S-461	Spray	1.2	15,20,neg.(2)	-
TCE/RH-195	Spray	2.7	5, neg.(3)	-
TCE/RH-195	Spray	0.7	9,30,20,neg.	30,neg(3)
TCE/RH-195	Spray	0.7	7,26,neg.(2)	-
TCE/RH-195	Spray	0.8	10,15,20,neg.	-

Note: Figures in parentheses indicate the number of tests which gave the same paper test readings.

3. Ingredients for the improved potassium oleate paste were obtained from the NDRC group at the DuPont Experimental Station. The potassium oleate suspension had the following composition by weight:

Tetrachloroethylene (Perclene)	67.3 parts
Potassium Oleate (28% water)	18.0 parts
Barium hydroxide octahydrate	2.75 parts
Aristowax 160/165	1.6 parts

To prepare the paste, 11.5 parts of S-210 were mixed with 100 parts of the above dispersion. Decontamination tests for H using the paste were made on 6" x 12" deck painted steel panels. The H contamination was approximately 20 g./yd^2. The tests for residual H are given in Table III.

TABLE III

Decontamination of H with Improved Potassium Oleate Paste

Amount Used Lb/yd^2	Paper Test Time (Minutes)		
	Immed.	3 Hrs.	24 Hrs.
0.8	4, 4	40, neg.	neg.
0.9	10, neg.	neg.	
0.9	neg.		
1.0	neg.		
1.1	neg.		
1.5	neg.		

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Conditions of Tests:

	Temperature	Rel. humidity
Vapor cup tests	80°F	51%
Liquid tests	78°F	56%

Information received from Lt. Connor by S. Moore.

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NAVY DEPARTMENT

Report on

A Controlled Laboratory Experiment to Compare Lesions Resulting from Application of Mustard, Lewisite, and Nitrogen Mustards to the Skin of the Forearms of Humans.

NAVAL RESEARCH LABORATORY
ANACOSTIA STATION
WASHINGTON, D. C.

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ABSTRACT

Vapor cups containing mustard, Lewisite, and nitrogen mustard, (HN_3), were employed to produce lesions of comparable size and severity on the forearms of humans. In two (2) other series of tests, drops of equal weight and drops of equal volume of each of the above vesicants were applied to the forearms of men to compare the vesicant powers of the agents and to study the type of lesion produced in each case. Color photographs, both still and motion picture, were taken throughout the experiment to show each lesion developing, reaching a peak and subsiding. To the best of our knowledge, no previous experiments have been performed by placing the three (3) basic types of gases (mustard, arsenicals, and nitrogen mustards) on the same individual, with systematic observation, including pictures, from the time of application of the gas through the development and healing of the lesion. It is felt that these observations will be an aid to other investigators who are either new in this field, or do not have the facilities for running this type of experiment.

AUTHORIZATION

1. This problem was authorized by Bureau of Ships letter, reference (a), and Bureau of Ships Project, reference (b).

Reference (a) BuShips S-S77-2(dz) Serial 811, of 17 December 1940.

Reference (b) BuShips Project No. 397/44 of 2 February 1944.

OBJECT OF TEST

2. The object of this test was to produce and study the different types of lesions caused by the application of varying amounts of the three (3) basic types of vesicant gases; i.e., mustards, arsenicals, and nitrogen mustards, to the skin of the forearm of humans.

DESCRIPTION OF MATERIAL UNDER TEST

3-1. The equipment used was the standard Edgewood-type vapor cups with (2) drops of liquid vesicant on the blotter paper at the base of the cup, and the drods developed by the Rockefeller Institute for Medical Research using a platinum needle, the end of which had been cut off at right angles to the shaft. The drods were calibrated so that it was possible to deliver one (1) milligram and five-tenths (0.5) microliter amounts of each gas for the weight and volume experiments respectively.

3-2. For the photographs, two (2) standard Speed Graphic cameras with F 3.5 and F 4.5 lenses and a 16 mm Ciné Kodak motion picture camera with an F 2.7 lens were employed. The film used was 4" by 5" Kodachrome for the stills and 16 mm Kodachrome for the motion pictures. Two (2) Photoflood #4 lamps furnished the illumination. A permanent background and base for the arms of the men were set up so that angles and lighting of the different pictures would be as near the same as possible. The subject's number and the time after application of the gas were marked on the forearm for each picture. For the still pictures, it was necessary to take three (3) exposures for each picture required in order to insure having the proper color effect. When the films were sorted and edited, the best of each set of three (3) were retained for the report. The motion picture film was cut and re-spliced to produce continuous pictures of each lesion developing and subsiding.

METHOD OF TEST

4. Because of the diversion of opinion of many investigators as to the type of lesion resulting from gases if applied by different methods, it was decided that the following procedure would be used:

(1) Gas applied by means of so-called vapor cup
(2) Gas applied in liquid form by equal weight (1 milligram)
(3) Gas applied in liquid form by equal volume (0.5 microliters)

4-1. Preliminary:

It was necessary to run preliminary screening tests to determine the proper times of application of

the cups and the amount by weight and volume of the liquid vesicant needed to produce a lesion of the desired size and severity:

(a) Vapor Cup Screening:

Cups, one (1) of H, two (2) of L, and one (1) of HN_3, set up as previously described, were applied to the forearms of one (1) man. Arbitrary times of ten (10) minutes for H, two (2) and six (6) minutes for L, and eighty (80) minutes for the HN_3 were chosen. After twenty-four (24) hours, the H showed a vesicle (V) 17 x 18 mm, the two-minute L numerous pinpoint vesicles (NPV) 15 x 16 mm, the six-minute L a vesicle (V) 22 x 20 mm, the HN_3 a vesicle (V) 15 x 13 mm. From these results, it was decided that for the actual experiment the cups should be applied ten (10) minutes for H, six (6) minutes for L, and one hundred (100) minutes for HN_3.

(b) Equal Weight Screening:

Two (2) men were used in order to find the proper amount of liquid gas to apply to produce a lesion of the required size. Comparing applications of five-tenths (0.5), one (1), two (2), and three (3) milligrams of mustard to the forearms of the two (2) men, it was found that one (1) milligram was the amount desired. The next step was to apply one (1) milligram of each agent to the forearm of one (1) man to see if the lesions from all vesicants would be satisfactory, and to obtain enough information to set up a tentative schedule for intervals between observations and photographs.

(c) Equal Volume Screening:

After the weight of one (1) milligram had been decided upon, a unit of volume of mustard comparable to one (1) milligram was chosen, this unit being five-tenths (0.5) microliters. A final screening test comparable to the one for equal weight was run by applying five-tenths (0.5) microliters of each vesicant to the forearm of one (1) man.

4-2. General:

From the results of the screening tests, it was decided to use twelve (12) men: six (6) for the vapor cup test, three (3) for the equal weight test, and three (3) for the equal volume test. The men had an application of each agent on the flexor surface of the left

forearm so that one (1) was at the mid-point, and the others equidistant from the mid-point and cubital fossa or wrist respectively. The position of application of each gas on the arm was rotated so that sensitivity of the different parts of the forearm could be determined and thus ruled out.

The lesions of the individuals were carefully guarded by keeping the men at leisure and using protective dressings on large vesicles. There was no unnecessary exposure to sunlight or active washing of the exposed area; swimming was prohibited. These steps were taken because it was desired to show a lesion developing, reaching a peak, and subsiding without interference by trauma, infection, etc.

The points particularly stressed in the early readings were as follows:

(1) First visible change of the contaminated area -- erythema, blanching, etc.

(2) Subjective symptoms -- itching, burning, etc.

During the entire experiment as many pictures as were deemed necessary by the medical officer were taken to provide a continuous series showing the most minute changes in the lesion.

Temperature and relative humidity were recorded throughout.

4-3. Experiment #1 - Vapor Cups.

As previously determined from screening tests, it was found that the order of severity of the vapor burns produced per unit time of application by the gases was as follows: Lewisite, mustard, nitrogen mustards. By calculation, it was determined that leaving the Lewisite cup on six (6) minutes, the mustard cup on ten (10) minutes, and the nitrogen mustard cup on one hundred (100) minutes would give approximately equally severe burns. Because it was the object of this experiment to produce burns in which any basic difference in characteristics could be noted and not difference in severity, the above

time limits for application of cups were used. It was necessary to use as zero (0) time in the experiment that time when the cups were removed. Therefore, the nitrogen mustard cup was applied first, in ninety (90) minutes the mustard cup, in ninety-four (94) minutes the Lewisite cup, and all were removed from the arm at the same time. Readings were taken every five (5) minutes up to two and one-fourth ($2\frac{1}{4}$) hours, every fifteen (15) minutes to four and one-fourth ($4\frac{1}{4}$)hours, every thirty (30) minutes to seven (7) hours, every hour to fourteen (14) hours, every two (2) hours to twenty-four (24) hours. Readings were taken at twenty-eight (28) hours, thirty-six (36) hours, and forty-eight (48) hours, with subsequent readings every twenty-four (24) hours until the lesions had subsided and healed.

(See Plates 1 - 6, 13 - 14)

4-4. Experiment #2 - Equal Weight.

Three (3) men had the gas in liquid form applied to the skin of their left forearms by the platinum pointed drod. The men were cautioned and watched until the gas had visibly disappeared not to move the forearms from a horizontal position. Therefore, any dispersion of gases on the skin was due to the character of the liquid and/or skin; i.e., capillary action, specific gravity, etc. Here it was particularly noted that:

(1) Gases applied on the skin area closest to the wrist immediately began to disperse in the skin folds, causing a linear type of burn. This was particularly true of nitrogen mustard, less so of Lewisite and mustard.

(2) Liquid gas on the other areas on the forearm remained as small circular droplets until complete evaporation took place except for a few small pseudopods which occasionally radiated out from these droplets.

In the application of the liquid gases, as much speed as possible was used so that the three (3) gases were applied almost simultaneously to the forearm. The average overall time of application was three (3) minutes. Any immediate visible reactions were looked for and any

METHOD OF TEST (Continued)

subjective symptoms noted. The same procedure for pictures and observations were followed as for Experiment 1.

(See Plates 7 - 9, 15 - 16)

4-5. Experiment 3 - Equal Volume:

The procedure for this experiment was exactly the same as for Experiment 2 except that the gases were applied by equal volume (0.5 microliters) instead of by equal weight. Again speed of application was stressed so that all three (3) gases were placed on the arm with a minimum of time difference -- average time of application was three (3) minutes.

(See Plates 10 - 12, 17 - 18)

DATA RECORDED DURING TEST: **(Plates 1 - 12)**

5-1. The progress of the lesions may best be shown by the use of two (2) graphs for each subject. The first graph shows the lesion developing to a peak, with an insert for the first four (4) hours; the second graph covers the complete reaction to the healing of the lesion.

5-2. In the graphs, no attempt was made to assign to the readings numerical values which would correspond directly to the severity of the lesion. Instead, the ascending and descending readings were merely placed on the abscissa of the graph in the order of their appearance.

5-3. The system used for reading the lesions approaching the peak was the routine one used at the Naval Research Laboratory with the insertion of four (4) additional readings which it was found advisable to use in this experiment.

(1) E -- Midway between an E- and an E.

(2) E+BP -- An area of E+ with a blanched papule in the center.

(3) NV -- Numerous small vesicles forming from an NPV which have not yet coalesced to form one (1) large vesicle.

(4) V Peak -- The vesicle at its maximum size.

DATA RECORDED DURING TEST (Continued)

5-4. The readings for the descending curve are the most common ones noted in the subsiding and healing lesions. They are self-explanatory.

5-5. For reading the graphs of the complete reaction, use the abscissa readings on the left for the ascending curve, the abscissa readings on the right for the descending curve.

5-6 An attempt was made to show the secondary reactions of Lewisite by a branching of the descending curve. The readings for the three (3) points on this branching are: papules periphery, vesicles periphery, and papules disappeared.

5-7. On all graphs, the solid line represents the mustard lesion, the dotted line the Lewisite lesion, and the broken line the nitrogen mustard lesion.

DISCUSSION

6. In this experiment, twelve (12) men were used; lesions were produced as previously described for each experiment. It was necessary to retain these men for five (5) weeks so that the lesions could be followed through to their conclusion. Certain definite characteristics were noted for each gas:

(1) Immediate reaction:

The first lesion to develop visibly was Lewisite, then nitrogen mustard, and finally mustard. The immediate erythema caused by Lewisite was much more pronounced than that caused by the other two (2) gases. It appeared rapidly and spread immediately over a wide area. It then rapidly diminished and assumed a normal size in comparison to the other two (2) lesions. This is a common characteristic of Lewisite burns and seems to be an easy way to identify this lesion. This may be classed as the immediate or primary reaction, but is no indication of the size of the final lesion as at least one-half ($\frac{1}{2}$) of the diameter disappears leaving no further sign of a burn at that site.

(2) Blanching:

Very close observation was maintained to determine if blanching appeared in the early states of Lewisite contamination. This phenomenon has been observed by other investigators and is said to consist of a blanching of the skin, appearing five (5) to fifteen (15)

minutes after Lewisite contamination. We were unable to collaborate these findings as can be seen in the colored photographs and movies. However, these blanching effects did occur later, not only in Lewisite burns but also in mustard burns.

(3) Vesicles:

The Lewisite vesicle was a tense lesion, and from the medical viewpoint could usually be classified as a so-called pedunculated type; i.e., the diameter of the base was as small as, or smaller, than the diameter of the mid-section. Also, the area of erythema surrounding the base of the vesicle was quite small. The mustard and nitrogen mustard vesicles could be classified as a sessile type of vesicle; i.e., the diameter of the base was larger than any other diameter. These vesicles also, in most cases, did not show the tenseness of the Lewisite lesions, and were surrounded by large haloes of erythema. The larger Lewisite vesicles in some cases appeared cloudy and opalescent. In these vesicles, after a few days, when they had begun to subside, the serum could be palpated as a clotted jelly-like mass.

No evidence of infection was found in the vesicles at any time as proved by bacteriological examination of the serum in the few cases that visibly appeared to be infected. The smears of the serum showed numerous polymorphonuclear leucocytes but no bacteria.

No attempt was made to drain the vesicles, but on the contrary, every attempt was made to preserve them and allow them to subside normally. In a few cases, they ruptured spontaneously, but most were preserved. In all except one (1) man, every burn resulted in the formation of a vesicle. This man was apparently highly resistant and could not be considered as a typical case.

(4) Severity of Lesions:

In the overall picture, the mustard lesions were the most pronounced. This was true not only for the diameter, but also for the depth of the lesion. Mustard lesions definitely injured the skin and caused, in most cases, third degree burns, exposing, and in some cases, damaging the corium. Lewisite lesions are next in severity

DISCUSSION (Continued)

severity, causing in many cases third degree burns, but usually the corium was spared. The nitrogen mustard lesions were comparatively superficial and resulted mostly in second degree burns, injuring the superficial and intermediate skin layers.

(5) Secondary Reactions:

The only secondary reactions observed were caused by Lewisite and occurred approximately fourteen (14) days after the original burn. These were evidenced by marked erythema around the original burn area, and the appearance of numerous small papules and vesicles surrounding the primary burn site. These small vesicles in the majority of the cases formed a halo around the original lesion. They were accompanied by intense itching but no pain.

(6) Pain:

Lewisite caused pain almost immediately after application. The mustard and nitrogen mustard lesions were practically painless. Any later discomforts were due to the lesions themselves and not to the causative agent.

(7) Position:

Those lesions at the center position on the forearm tended to peak more rapidly than those at the other two (2) positions. However, the final size and severity of the lesions were not significantly affected by the position of the lesion on the arm except that the lesions at the wrist were invariably of a linear type.

RESULTS

7. (1) Immediate itching and burning appeared with Lewisite, occasional itching with nitrogen-mustard, and no immediate symptoms with mustard.

(2) The mustard and Lewisite vesicles were approximately the same in size. The nitrogen mustard vesicles were generally smaller.

(3) The average lengths of time required by the lesions to reach a peak (maximum vesication) were thirty-nine (39) hours for mustard and forty-two (42) hours for Lewisite and nitrogen mustard.

RESULTS (Continued)

(4) In comparing lesions by weight with lesion by volume, the nitrogen mustard by weight were larger than the nitrogen mustard by volume. No significant difference was observed between the mustard weight and volume lesions or the Lewisite weight and volume lesions.

(5) All three (3) agents produced a linear type of lesion when applied to the wrist areas.

(6) Those lesions whose roofs remained intact, although they may previously have been ruptured, healed more quickly than those lesions which became denuded, and formed a true scab later.

(7) The time of healing of the lesions was as follows: nitrogen mustard first, Lewisite second, and mustard last, irrespective of the secondary reactions from Lewisite.

(8) The nitrogen mustard lesions showed the most induration (edema).

(9) Pigmentation of the skin appeared first in nitrogen mustard, then mustard, and last Lewisite. Pigmentation was most pronounced in nitrogen mustard, less in mustard, and least in Lewisite.

(10) Varying degrees of concavity were observed in the mustard and nitrogen mustard lesions after the formation of scabs, the mustard lesions being the most pronounced.

(11) The secondary reactions from Lewisite appeared about fourteen (14) days after the onset of the initial lesion. Thirty-one (31) days later these reactions were still evident. No secondary reactions appeared with mustard and nitrogen mustard.

(12) The final scars were most severe with mustard, less with Lewisite, and least with nitrogen mustard.

(13) No constitutional symptoms could be demonstrated at any time by routine physical examinations and laboratory procedures.

CONCLUSIONS

8. (1) Mustard gas causes the most severe lesions, Lewisite is next in severity, and nitrogen mustard the least. The healing time is in the reverse order.

(2) The average peak lesion is reached in thirty-nine (39) hours with mustard, and in forty-two (42) hours with Lewisite and nitrogen mustard.

(3) The Lewisite vesicle can be distinguished from the mustard and nitrogen mustard vesicles by its pedunculated appearance and the tendency of the serum to coagulate within the vesicle.

(4) The nitrogen mustard lesions produced by weight (1 milligram of the agent) are larger than those produced by volume (0.5 microliters of the agent).

(5) Pigmentation of the skin appears first and is most pronounced in the nitrogen mustard lesions.

(6) Lewisite causes an immediate erythema over a wide area which rapidly diminishes and assumes a normal size in comparison to the other lesions.

(7) Those lesions whose roofs remain intact heal more quickly than those which become denuded and form a true scab later.

(8) The center position on the forearm is slightly more sensitive than the other two (2) positions. Those lesions at the wrist position are linear in character.

PLATE INDEX

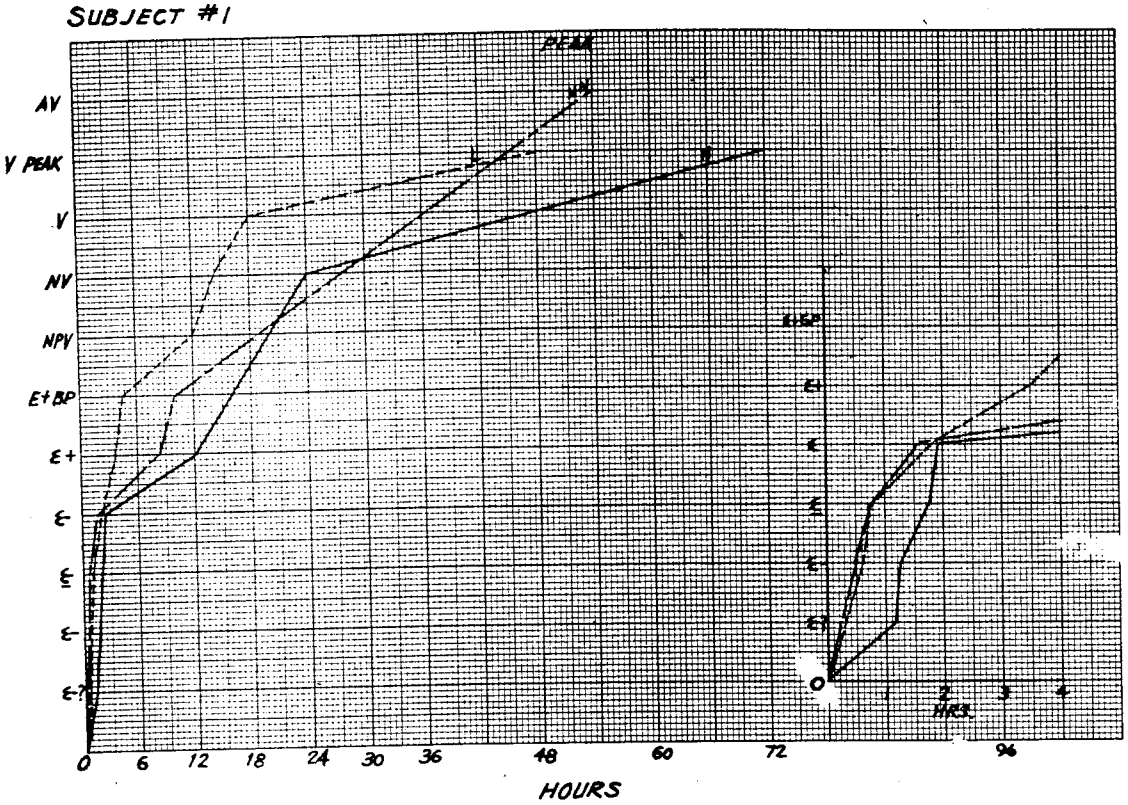
1 - 6	Experiment #1.	Vapor Cup Test	Graphs
7 - 9	Experiment #2.	Equal Weight Test	Graphs
10 - 12	Experiment #3.	Equal Volume Test	Graphs

For explanation of graphs see page #6.

13 - 14	Experiment #1. Subject #[redacted]1	Vapor Cup Test	Pictures
15 - 16	Experiment #2 Subject #9	Equal Weight Test	Pictures
17 - 18	Experiment #3 Subject #11	Equal Volume Test	Pictures

Pictures show a representative man from each experiment.

SUBJECT #1



EXP #1
SUBJECT #1

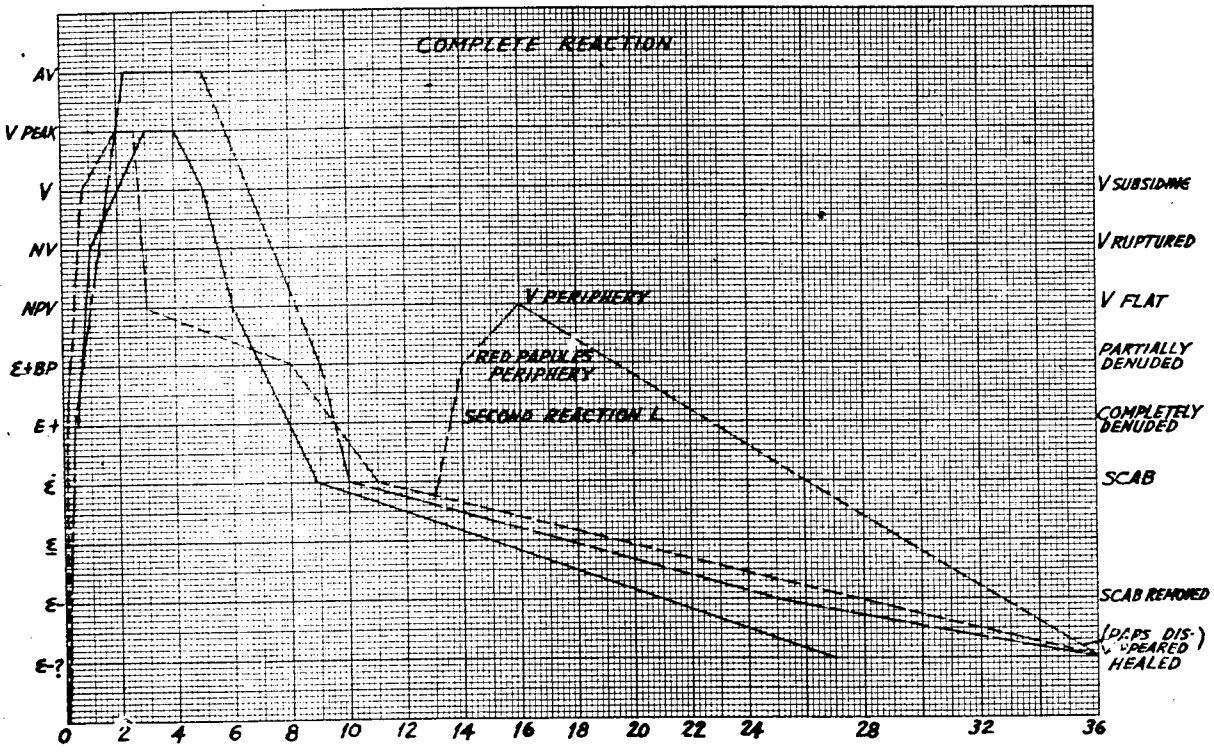
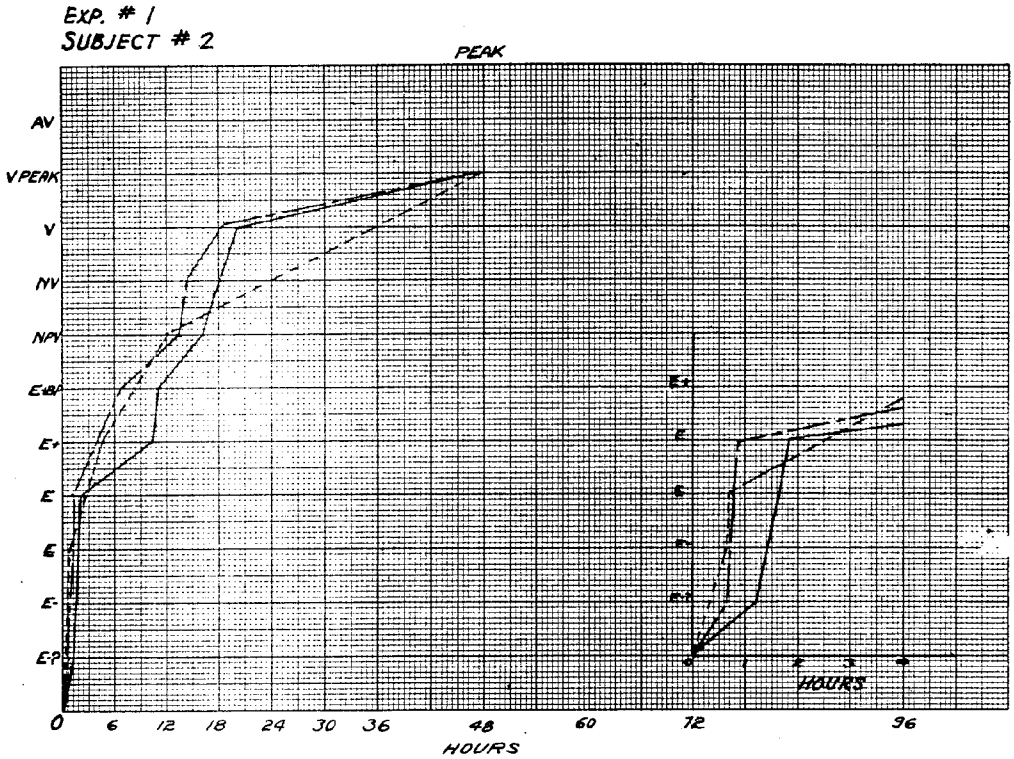


PLATE I


EXP. # 1
SUBJECT # 2
PEAK
AV
V PEAK
V
NV
NPV
E+BP
E+
E
E
E-
E-?
0
6
12
18
24
30
36
48
60
72
96
HOURS
HOURS

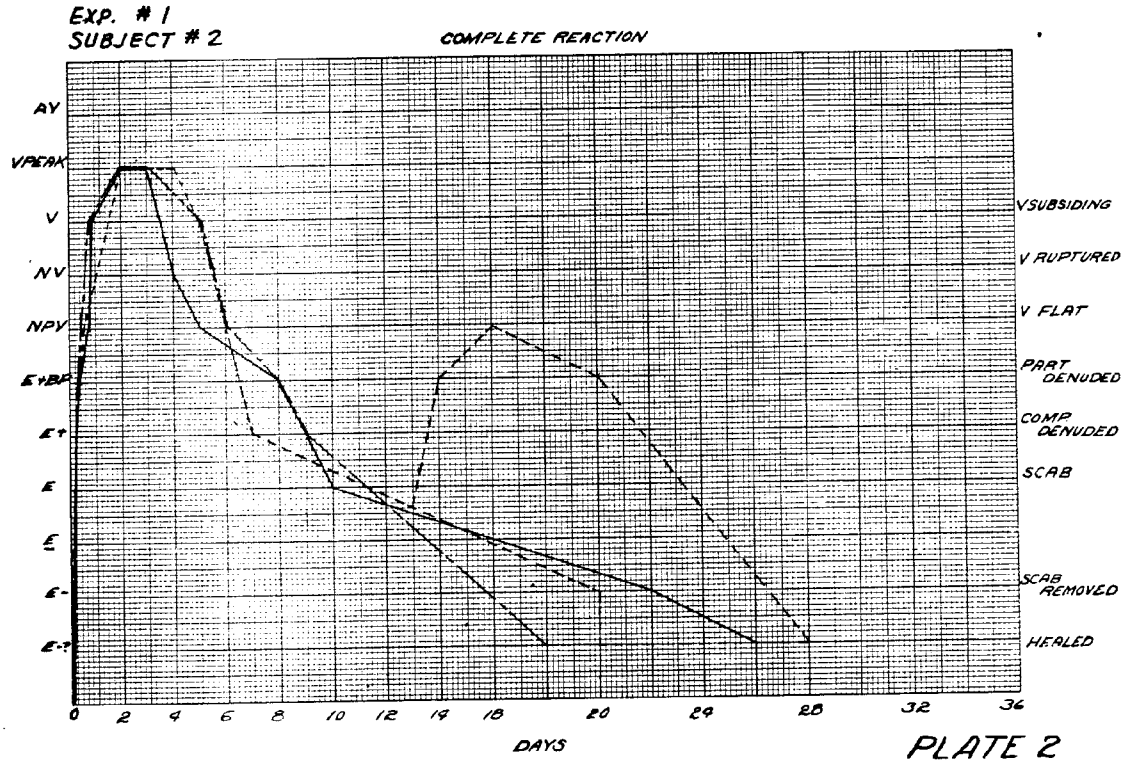
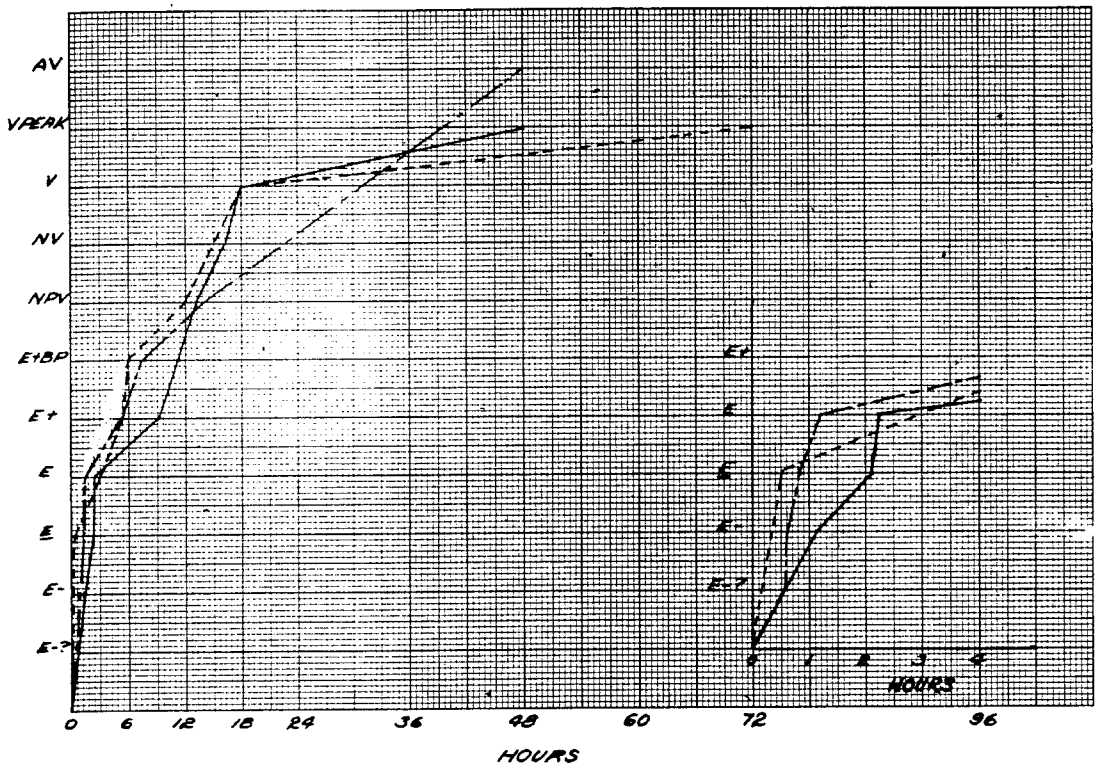

EXP. # 1
SUBJECT # 2
COMPLETE REACTION
AV
V PEAK
V
NV
NPV
E+BP
E+
E
E
E-
E-?
V SUBSIDING
V RUPTURED
V FLAT
PART DENUDED
COMP DENUDED
SCAB
SCAB REMOVED
HEALED
0
2
4
6
8
10
12
14
18
20
24
28
32
36
DAYS

PLATE 2

EXP. # 1
SUBJECT # 3

PEAK



EXP. # 1
SUBJECT # 3

COMPLETE REACTION

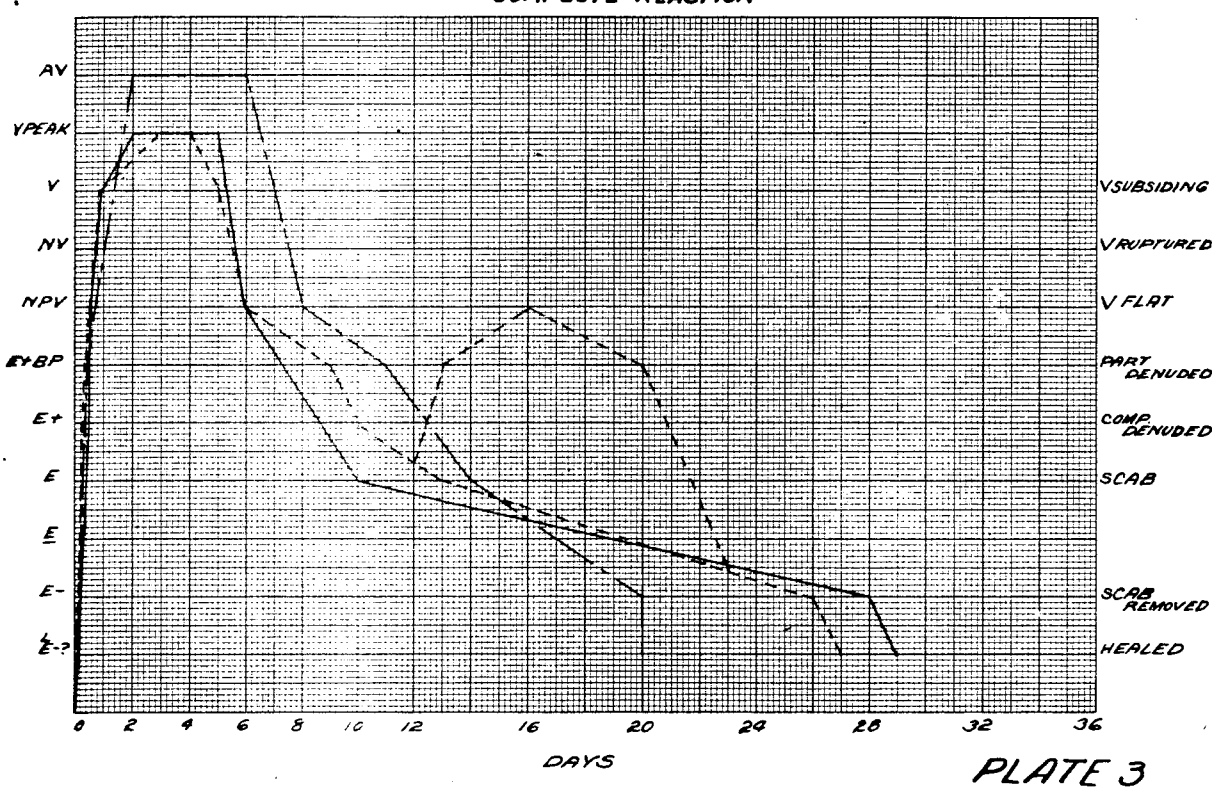
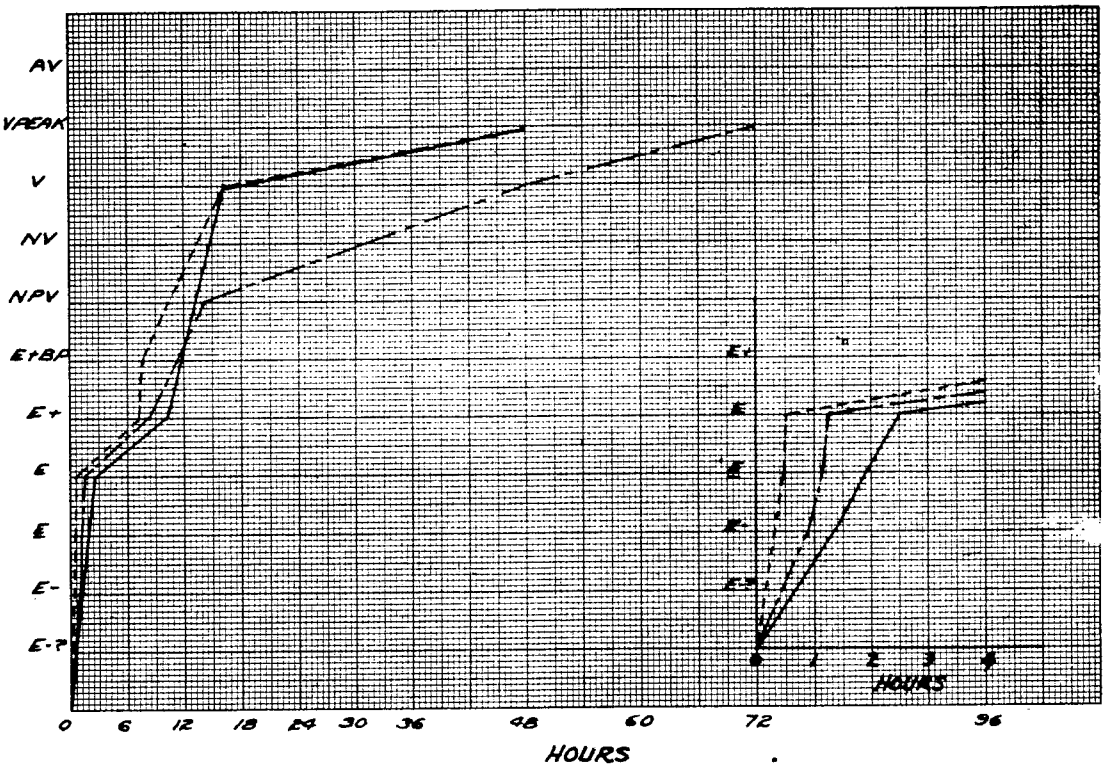


PLATE 3

EXP. # 1
SUBJECT # 4

PEAK



EXP. # 1
SUBJECT # 4

COMPLETE REACTION

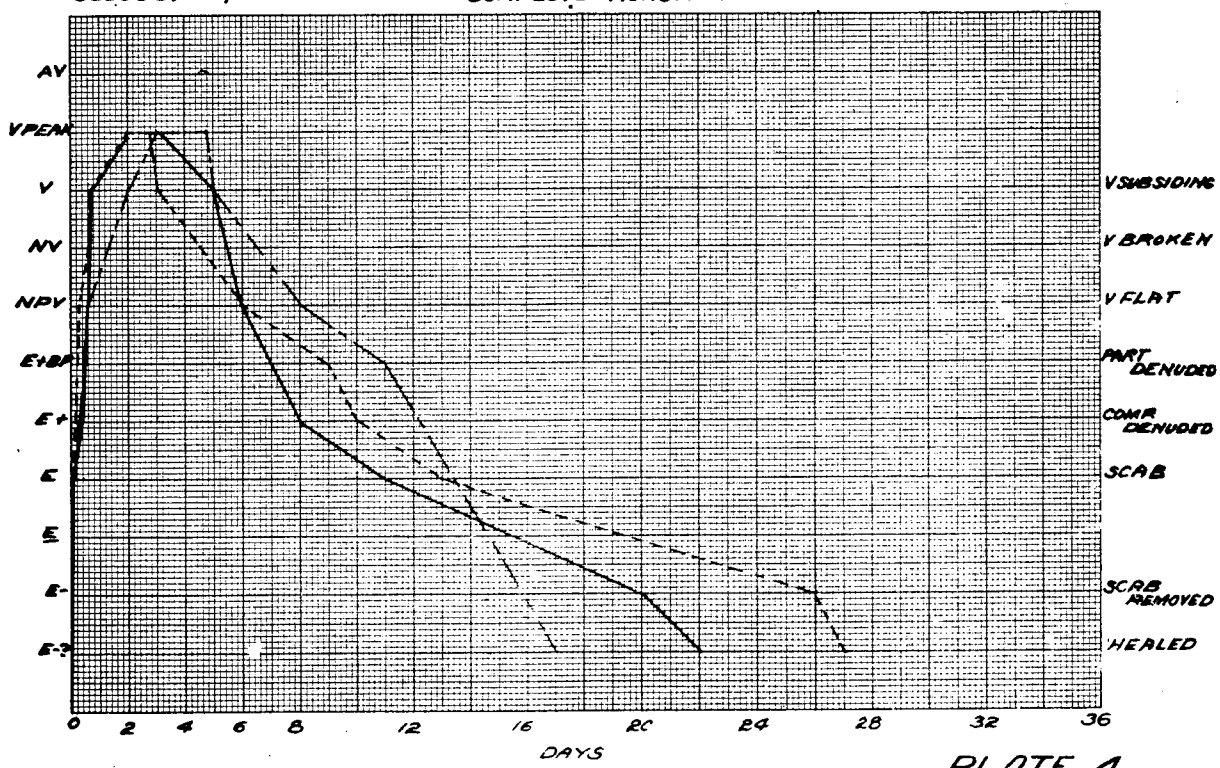
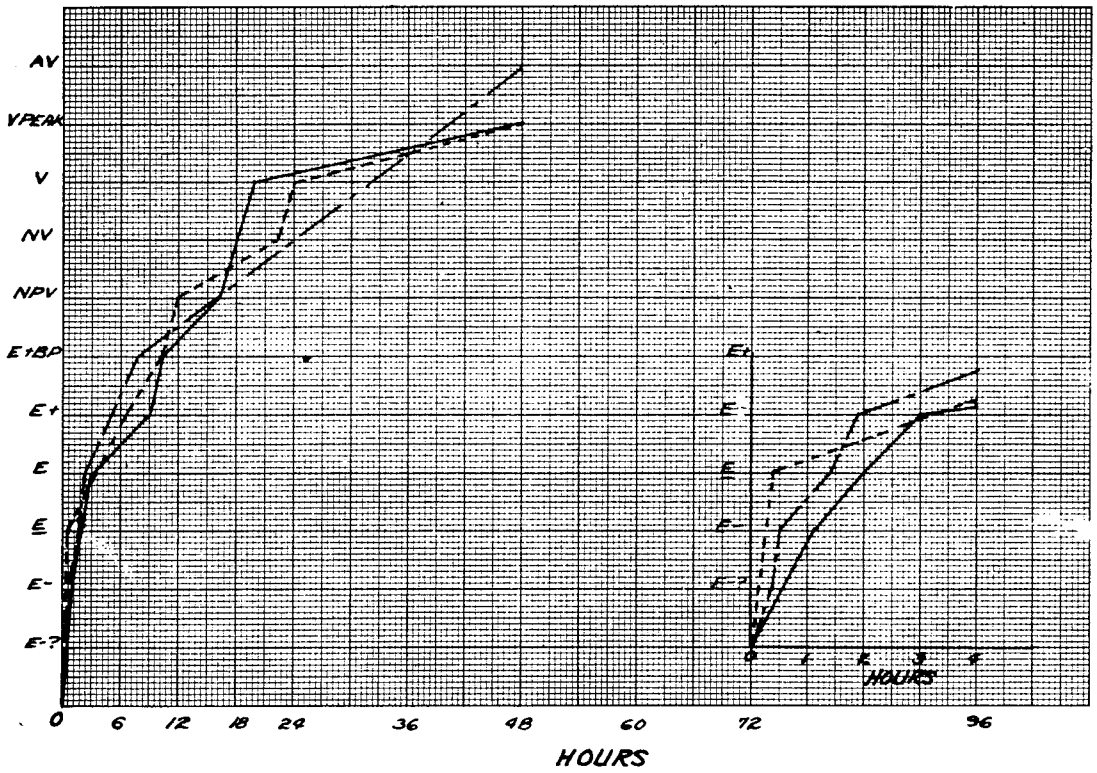


PLATE 4

EXP. # 1
SUBJECT #5

COMPLETE REACTION



EXP. # 1
SUBJECT #5

COMPLETE REACTION

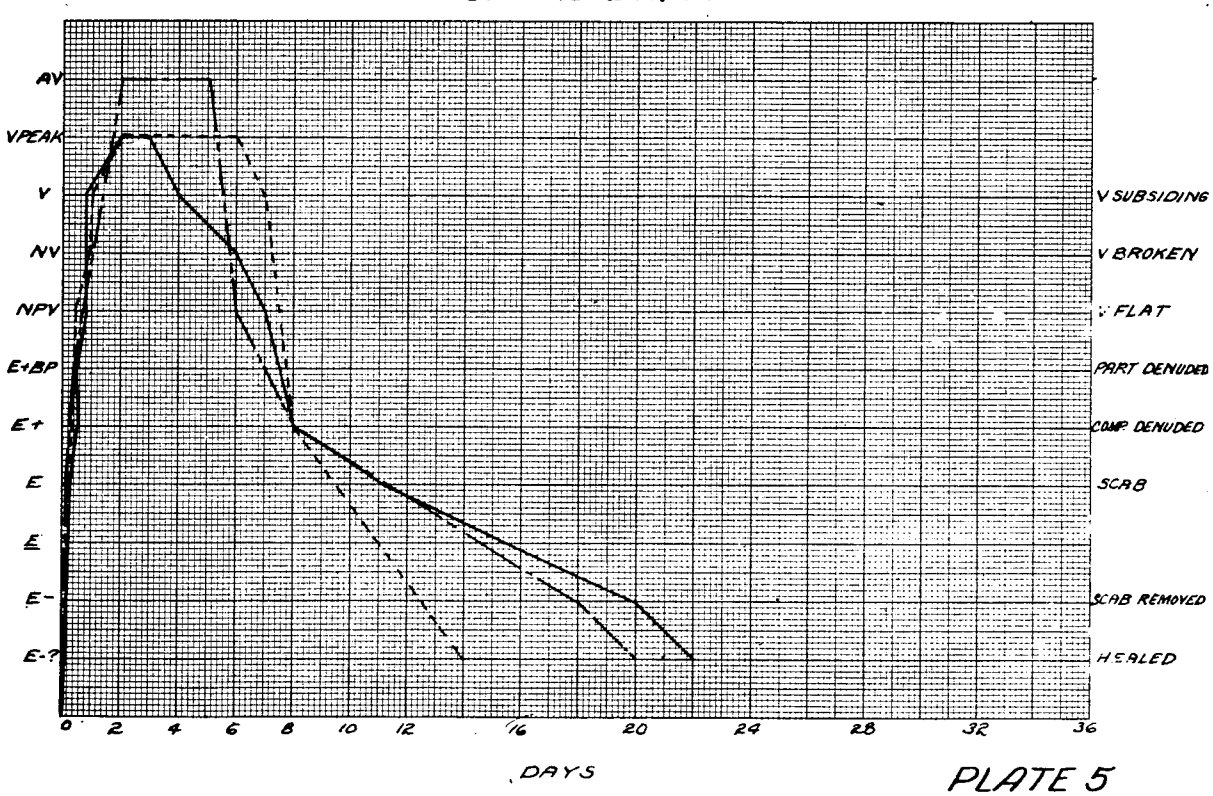
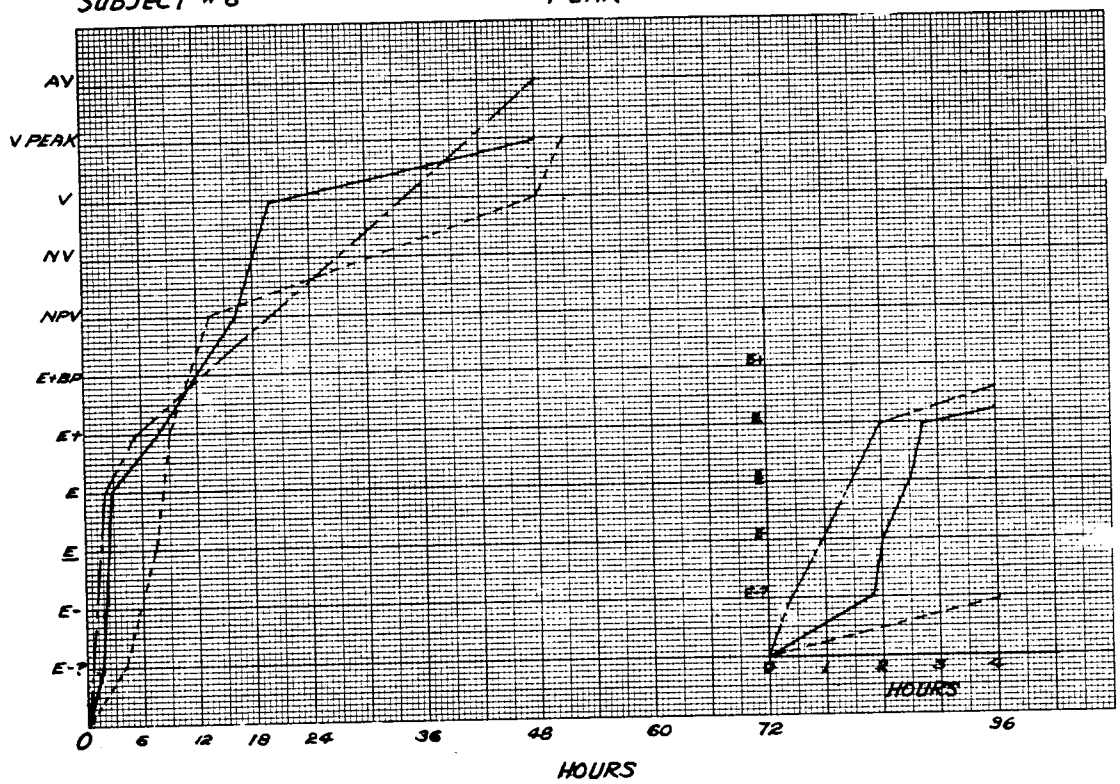


PLATE 5

EXP. #1
SUBJECT #6

PEAK



EXP. #1
SUBJECT #6

COMPLETE REACTION

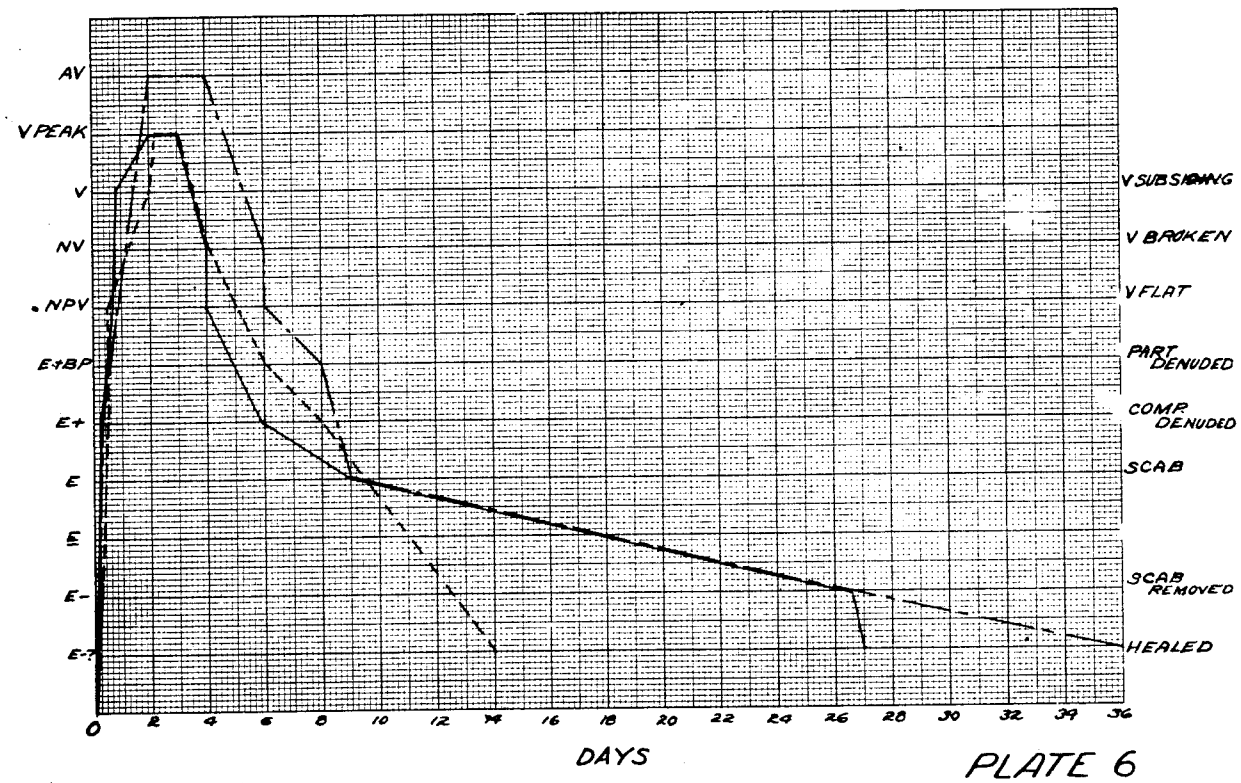
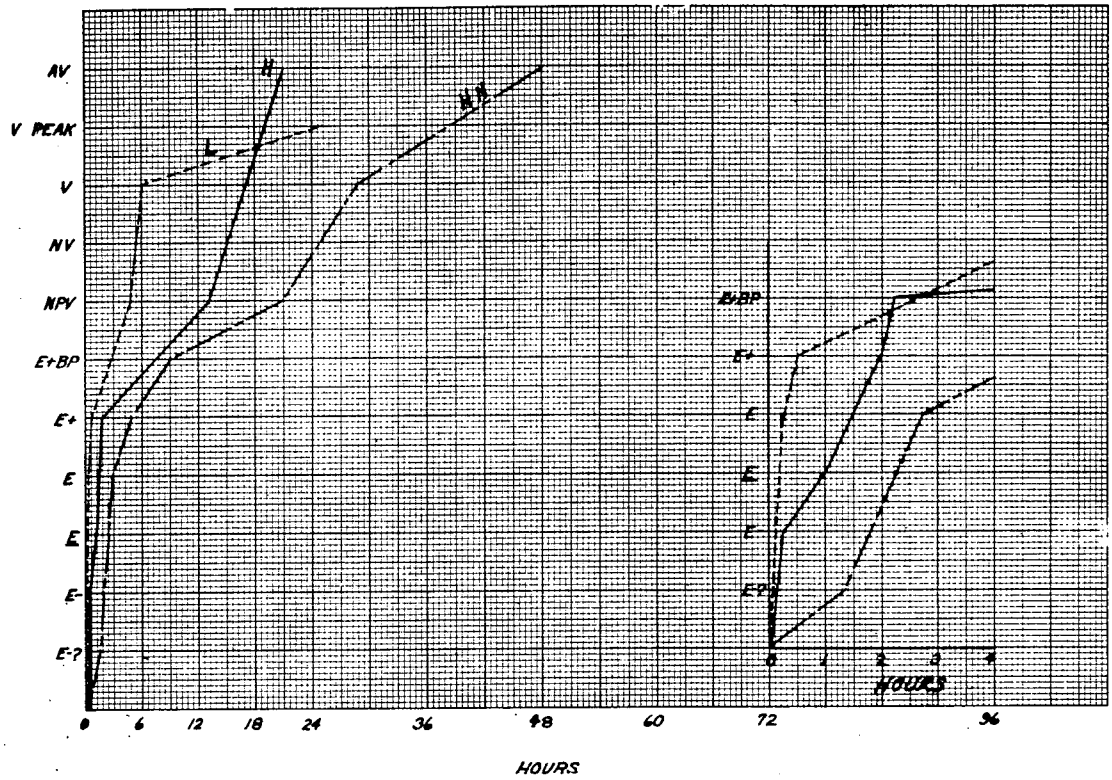


PLATE 6

EXP. # 2
SUBJECT #7

PEAK



EXP. # 2
SUBJECT # 7

COMPLETE REACTION

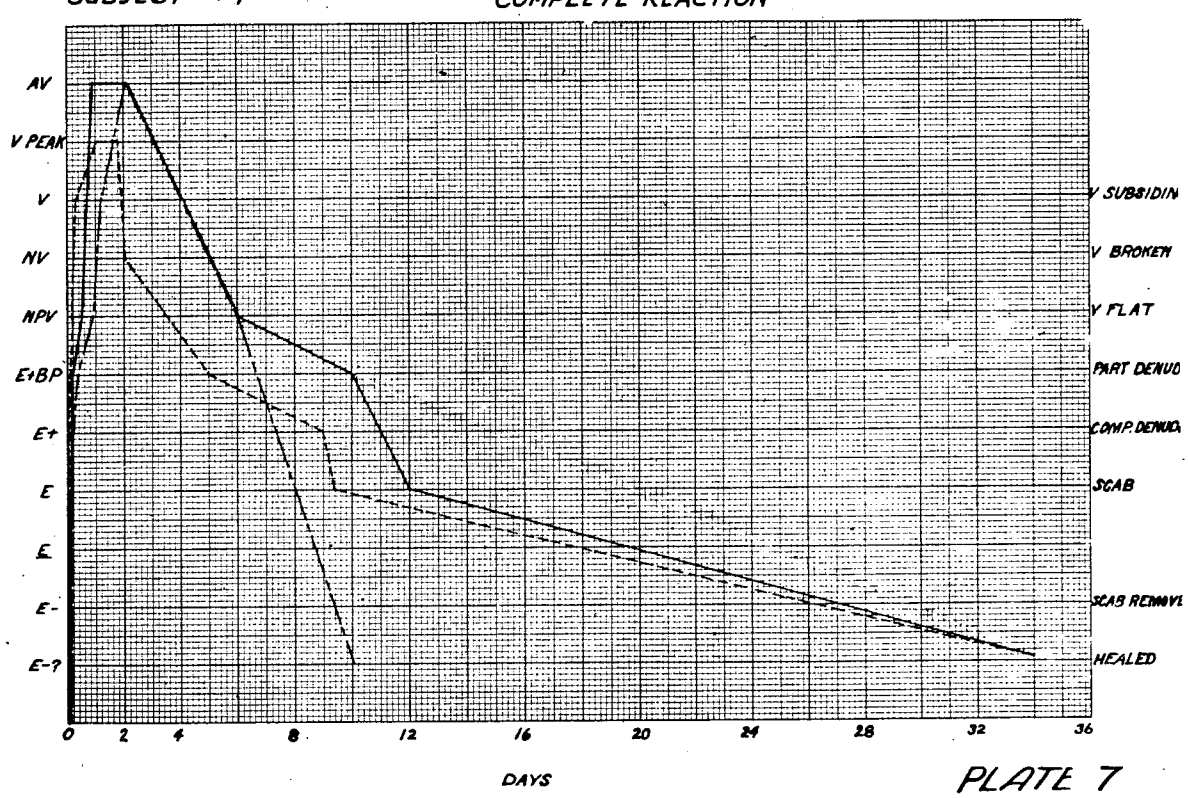
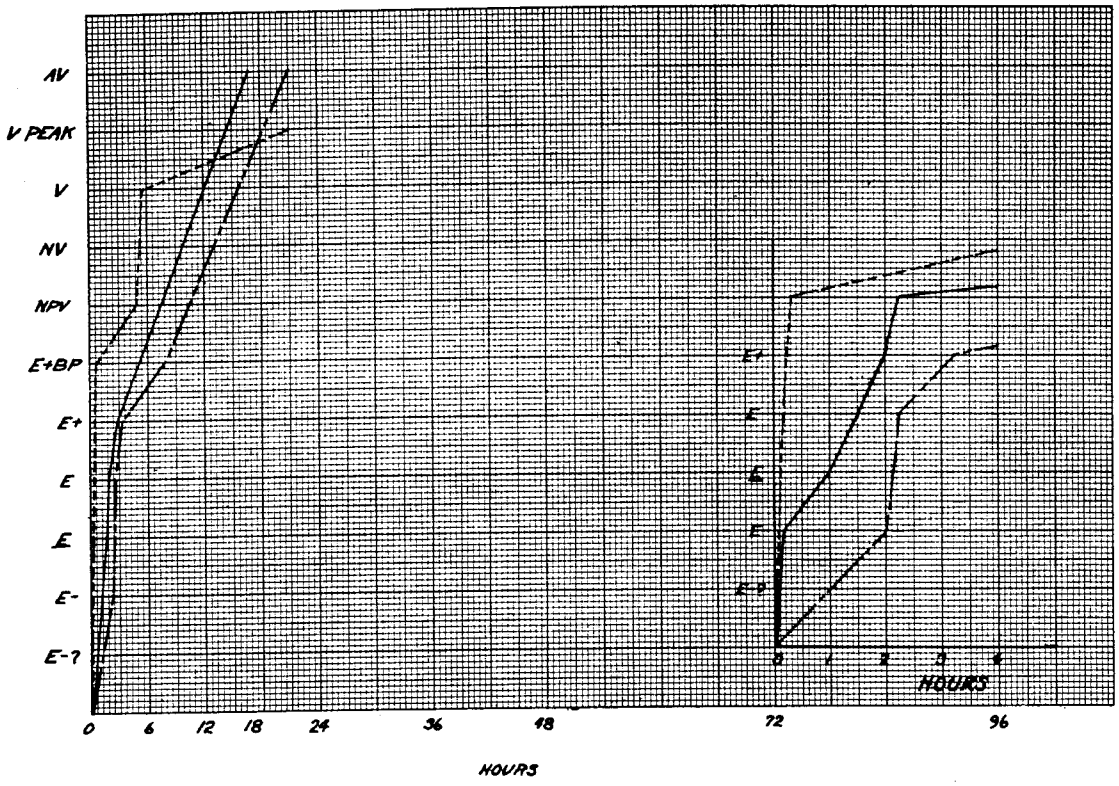


PLATE 7

EXP. # 2
SUBJECT # 8

PEAK



EXP. # 2
SUBJECT # 8

COMPLETE REACTION

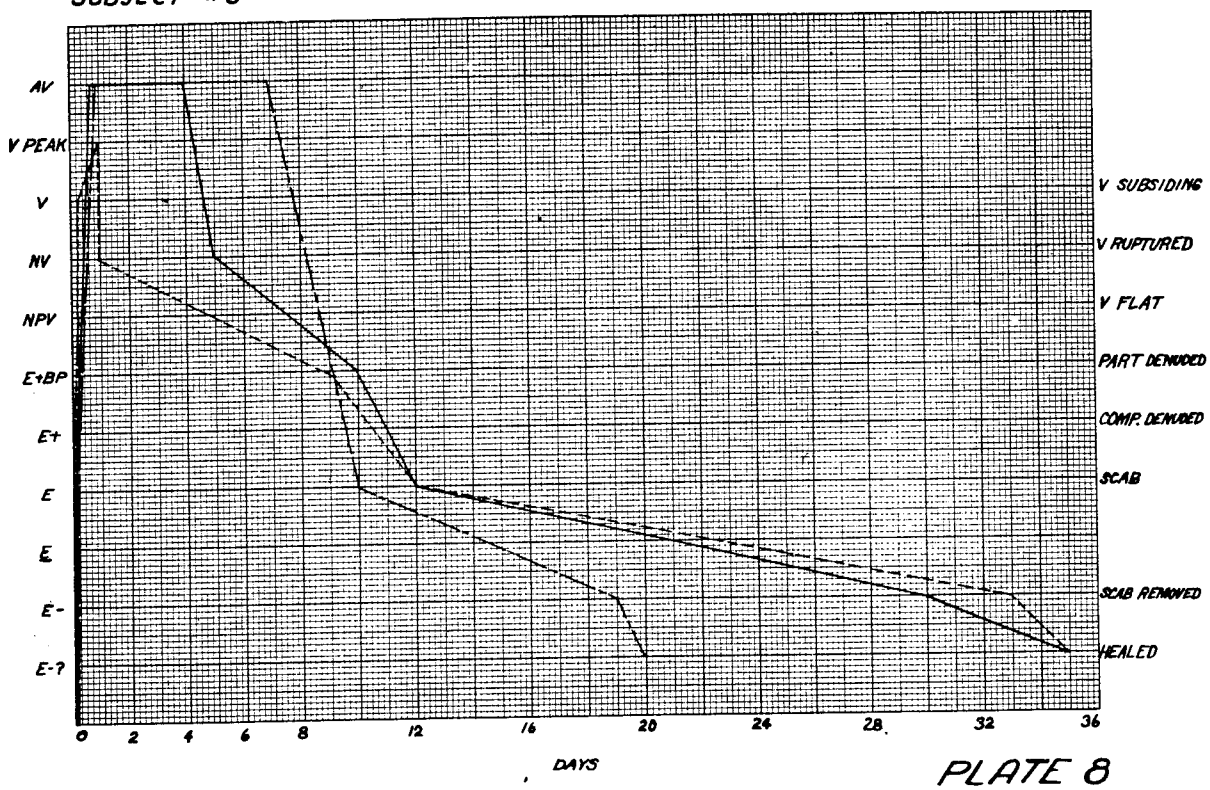
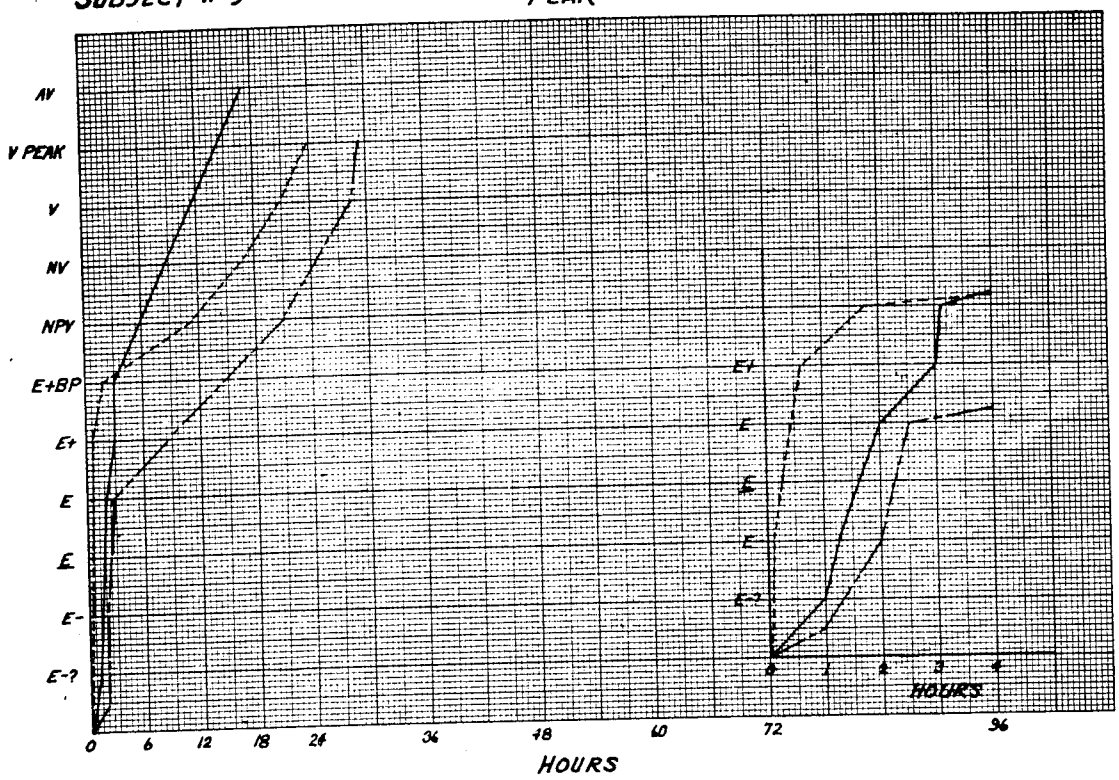


PLATE 8

EXP. # 2
SUBJECT # 9

PEAK



EXP. #2
SUBJECT #9

COMPLETE REACTION

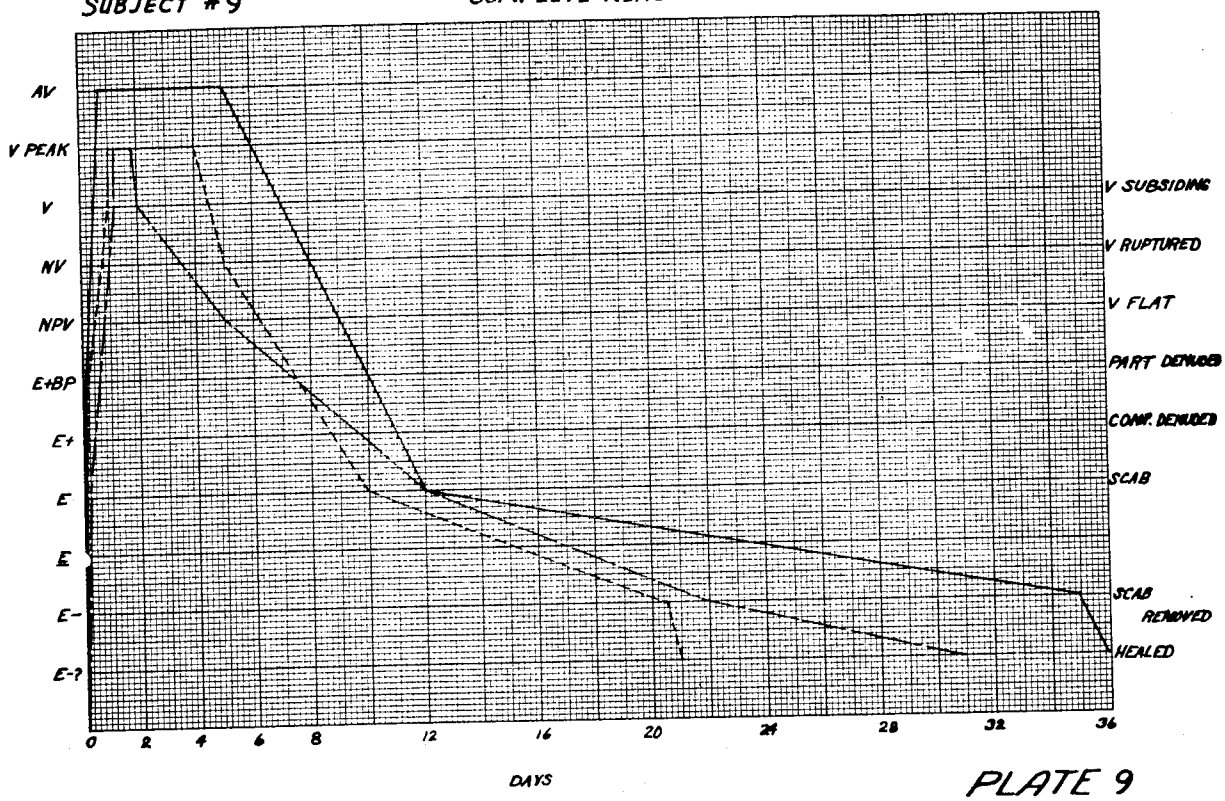
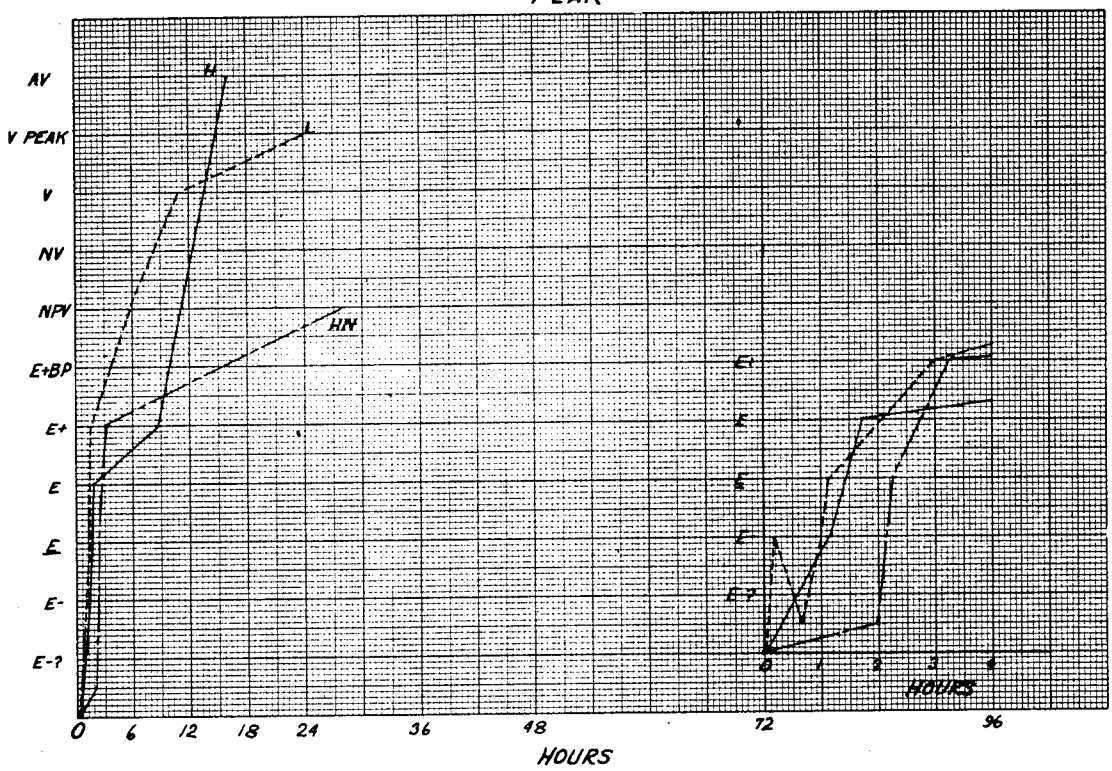


PLATE 9

EXP. # 3
SUBJECT # 10

PEAK



EXP. # 3
SUBJECT # 10

COMPLETE REACTION

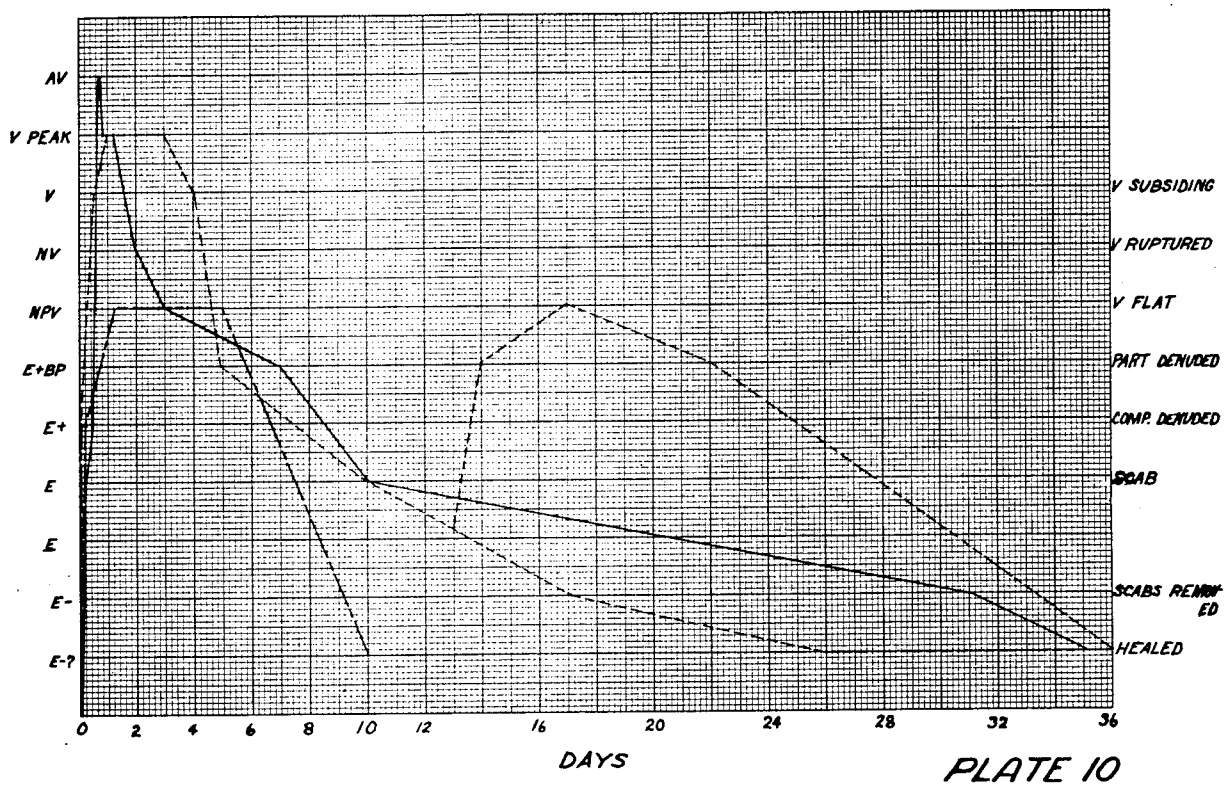
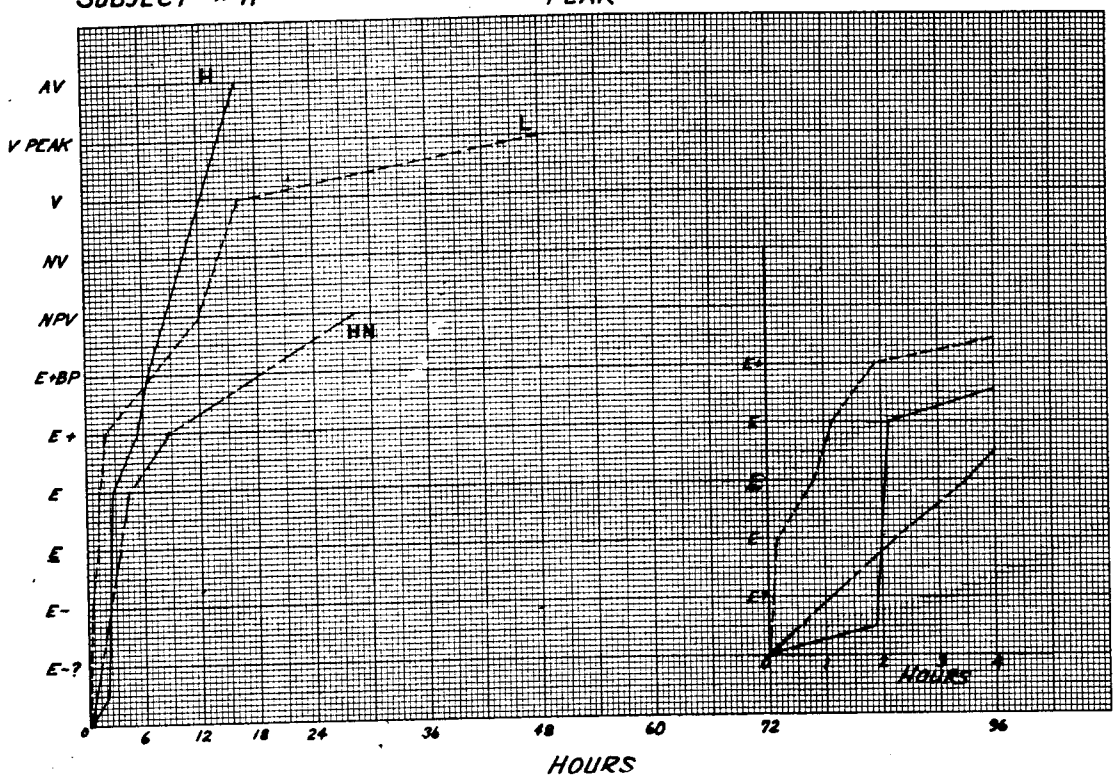


PLATE 10

EXP. # 3
SUBJECT # 11

PEAK



EXP. # 3
SUBJECT # 11

COMPLETE REACTION

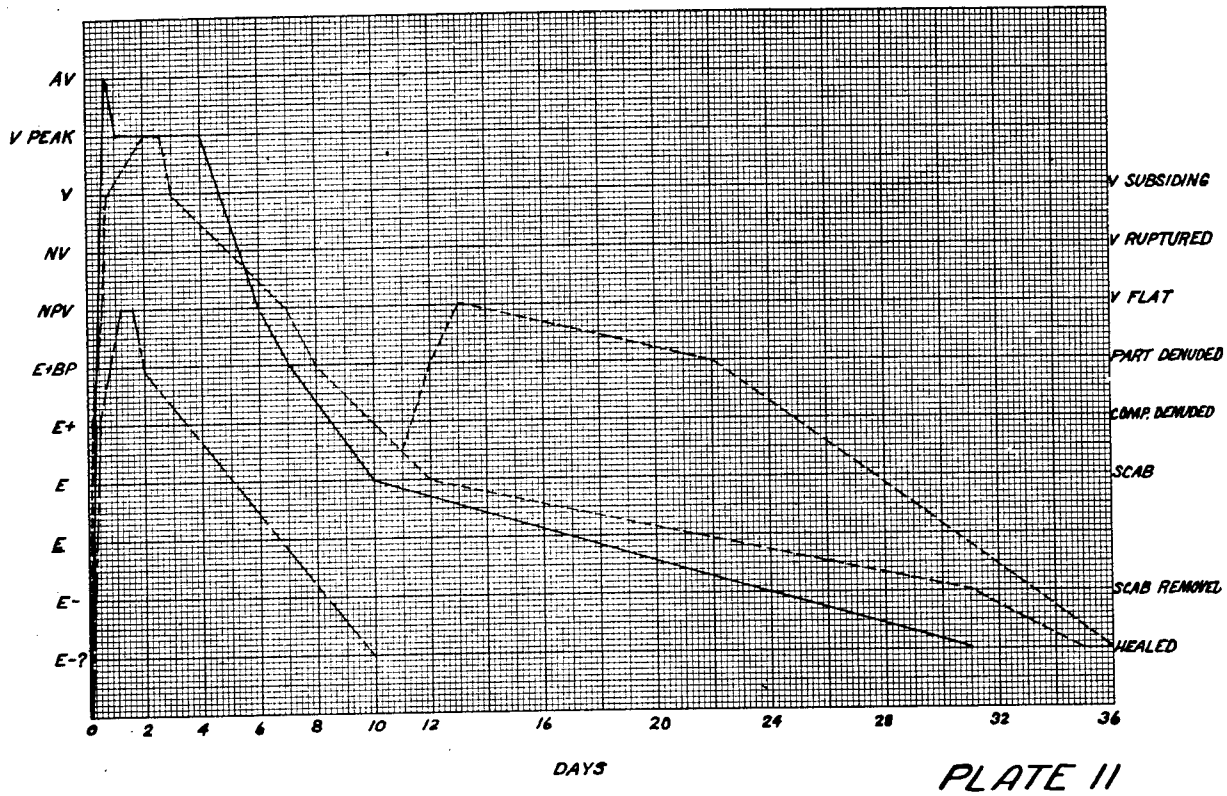
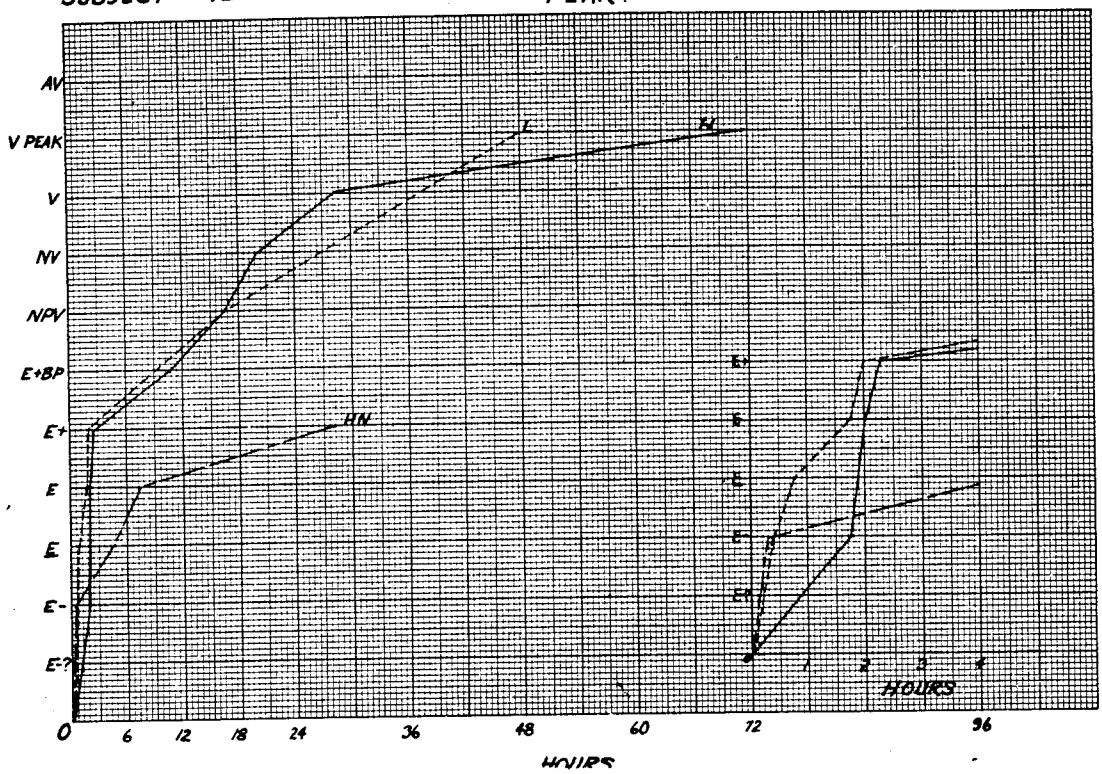


PLATE 11

EXP. # 3
SUBJECT #12

PEAK



EXP. # 3
SUBJECT # 12

COMPLETE REACTION

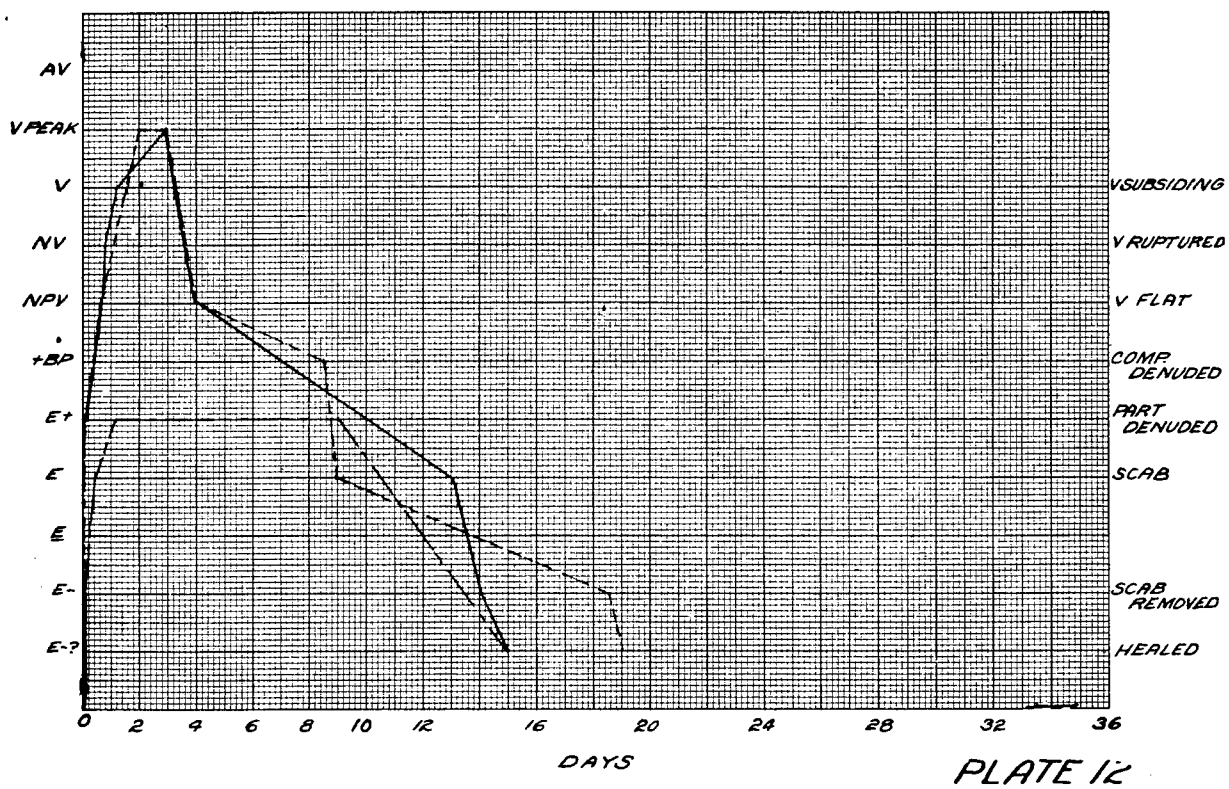
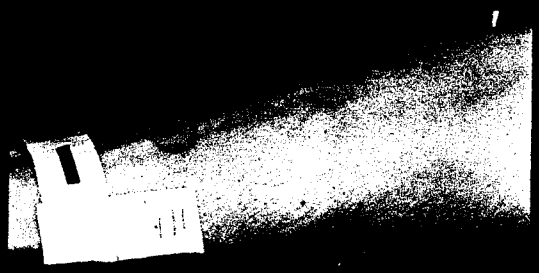


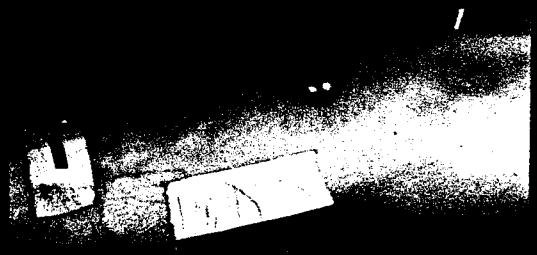
PLATE 12

EXPERIMENT NO. 1 REPRESENTATIVE MAN USING VAPOR CUPS. MAN NO. 1

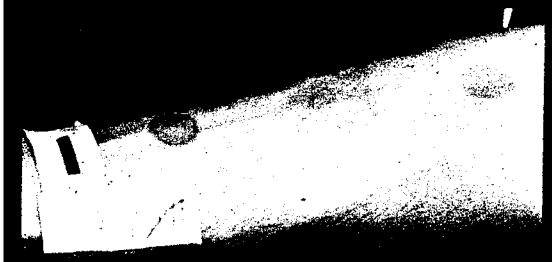
H (PROXIMAL) L (MIDDLE) HN (DISTAL)



1½ Hours

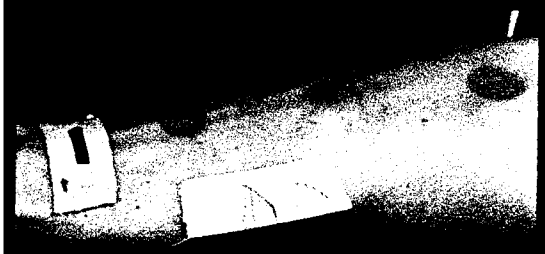


24 Hours



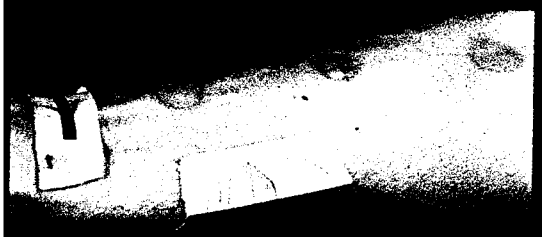
4 Hours

36 Hours



8 Hours

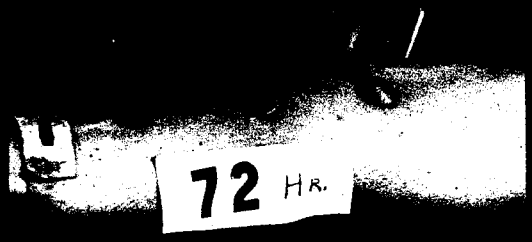
48 Hours



20 Hours

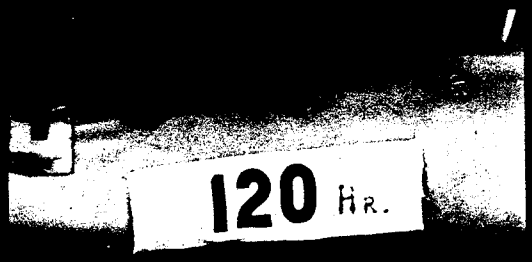
54 Hours

Plate 13



72 Hours

13 Days



120 Hours

16 Days

144 Hours

21 Days

8 Days

27 Days

Plate 14

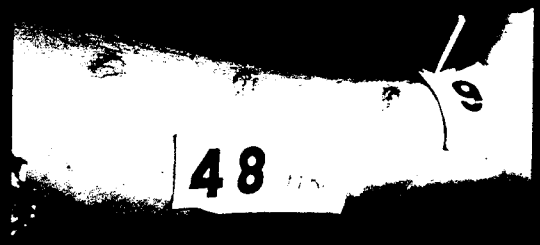
REPRESENTATIVE MAN IN EXPERIMENT NO. 2 USING WEIGHT. MAN NO. 9

HN (PROXIMAL) H (MIDDLE) L (DISTAL)

2½ Hours

28 Hours

8 Hours

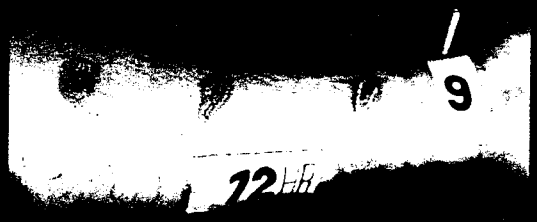


48 Hours

10 Hours

72 Hours

21 Hours



72 Hours

Plate 15

96 Hours

12 Days

120 Hours

15 Days

7 Days

20 Days

8 Days

23 Days

Plate 16.

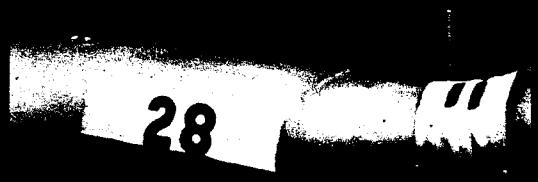
REPRESENTATIVE MAN IN EXPERIMENT NO. 3 USING VOLUME. MAN NO. 11

L (PROXIMAL) HN (MIDDLE) H (DISTAL)

2 Hours

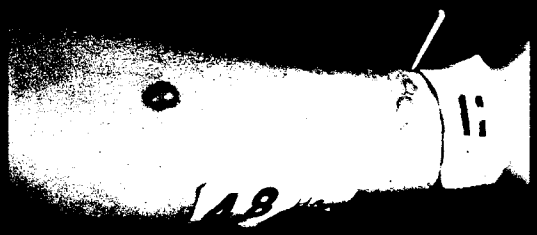
20 Hours

6 Hours



28 Hours

9 Hours



48 Hours

16 Hours

72 Hours

Plate 17

96 Hours

10 Days

120 Hours

15 Days

144 Hours

20 Days

7 Days

23 Days

Plate 18